

Sensing through Signal Amplification



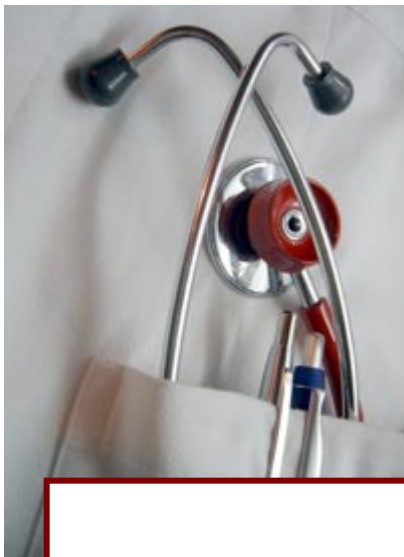
Leonard Prins

University of Padova

leonard.prins@unipd.it



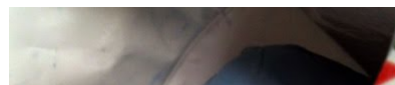
The importance of detecting ultralow (bio)chemical concentrations



Disease diagnosis



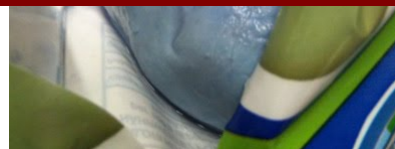
DNA diagnostics



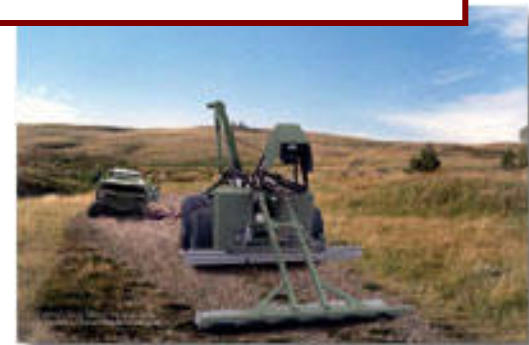
The ultimate detection limit
'Naked eye detection of a single molecule in a complex mixture'



Environmental pollution



Food contamination



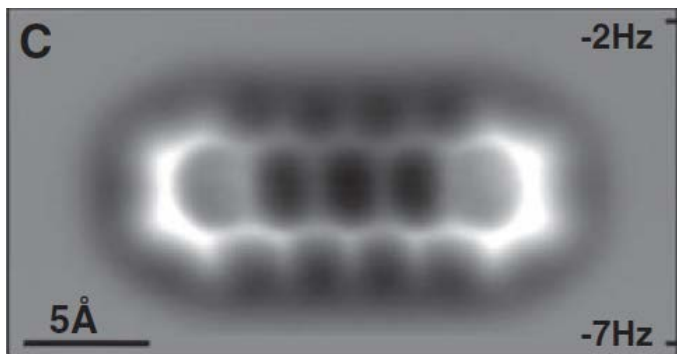
Explosives detection

..and more



Single molecule detection

TEM, SEM, AFM
SPR, SERS, X-ray



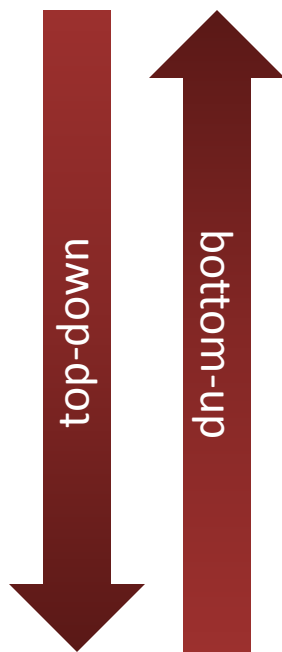
pentacene

Gross *et al. Science*, **2009**, 325, 1110-1114

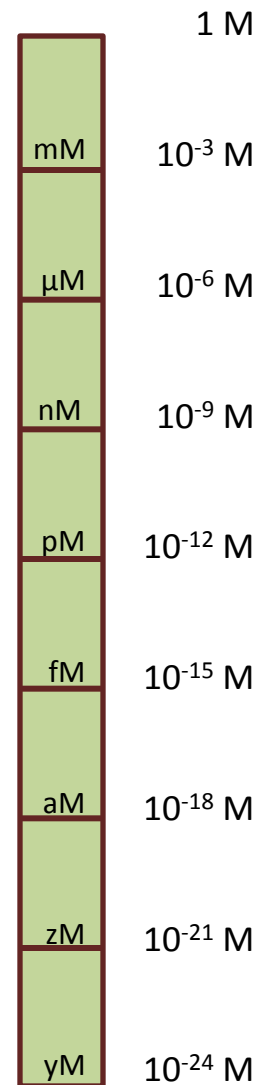
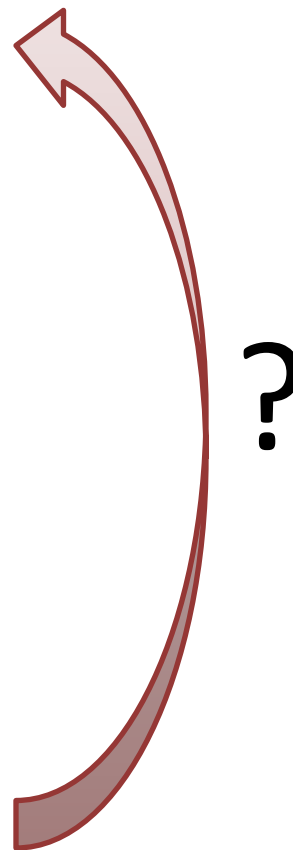


DNA smiley

Rothemund, *Nature*, **2006**, 440, 297-302



single molecule



$$1 \text{ molecule} / 1 \text{ liter} = 1.66 \times 10^{-24} \text{ M} = 1.66 \text{ yM}$$



Earth
≈
 5.97×10^{24} kg



=
1 kg



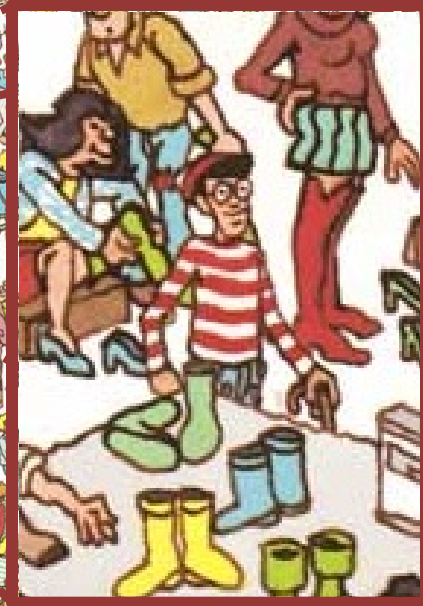
Where is Wally ?

HEY, WALDO-WATCHERS!
SAW SOME TRULY TERRIFIC
SIGHTS TODAY—SOMEONE
BURNING TROUSERS WITH
AN IRON; A LONG THIN MAN
WITH A LONG THIN TIE;
A GLOVE ATTACKING A MAN.
PHEW! INCREDIBLE!

Waldo



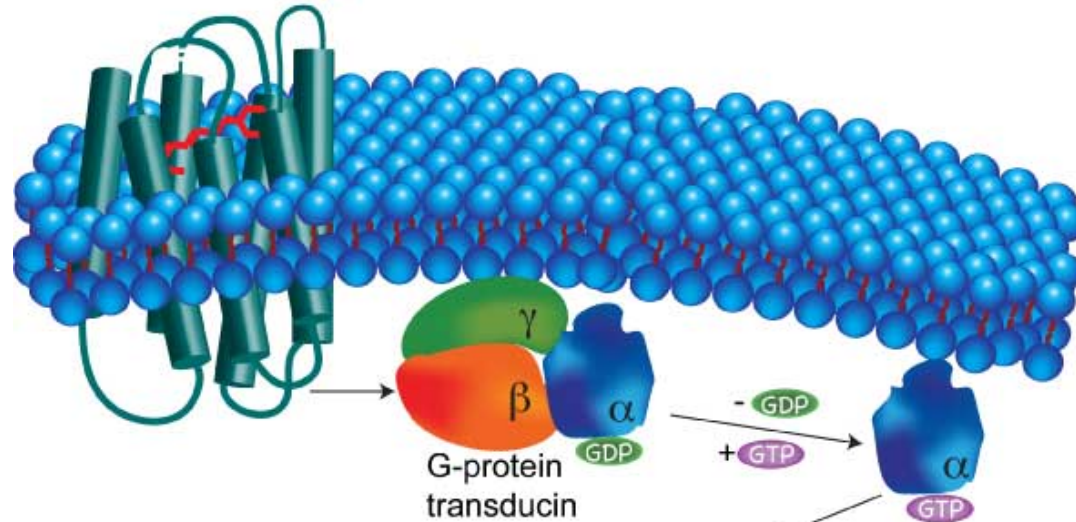
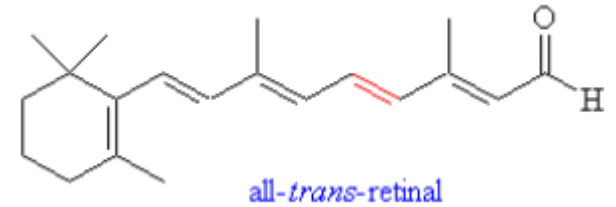
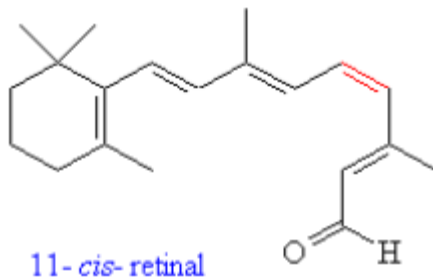
TO:
WALDO-WATCHERS
OVER THE MOON,
THE WILD WEST,
NOW





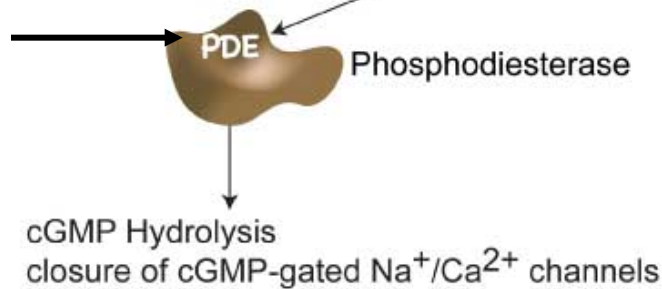
Signal amplification in Nature - catalysis

vertebrate visual response



- Light hits a rod cell and isomerizes retinal.
- ↓
- Rhodopsin converts to metarhodopsin II.
- ↓
- Metarhodopsin II activates transducin.
- ↓
- Transducin activates phosphodiesterase.
- ↓
- Phosphodiesterase hydrolyzes cyclic GMP.
- ↓
- Cyclic GMP is scarce, so Na⁺ channels close.
- ↓
- The membrane is hyperpolarized.
- ↓
- An electrical impulse is sent to the brain.

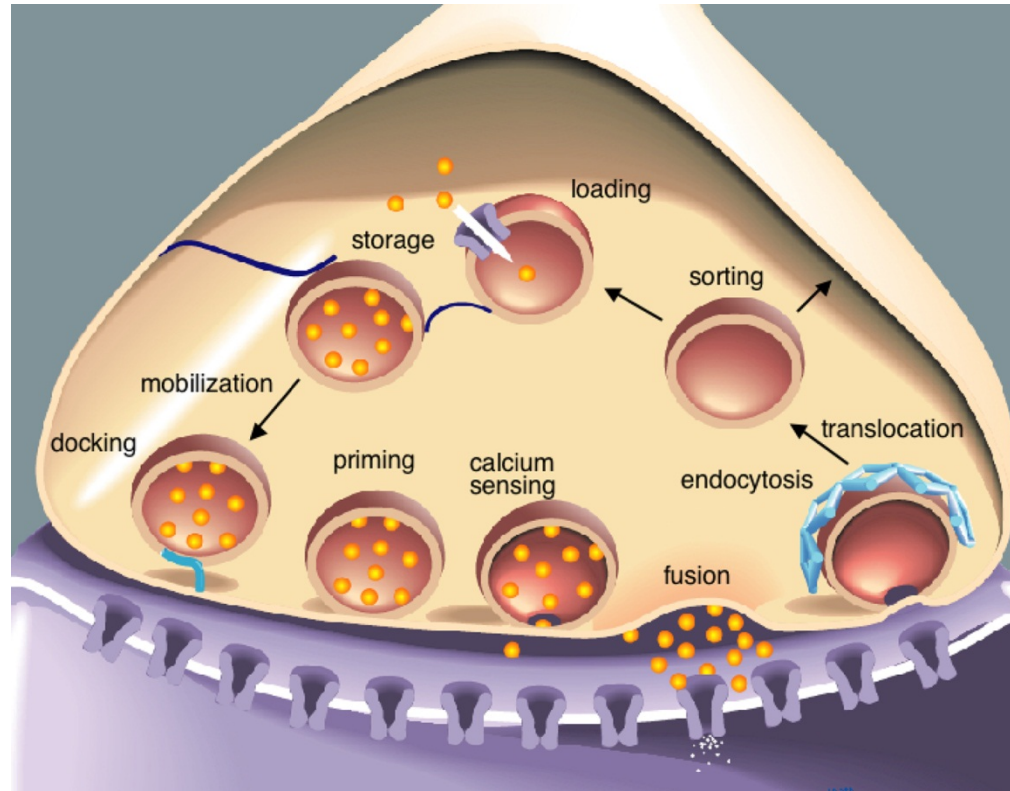
$k_{\text{cat}} = 4200 \text{ s}^{-1}$



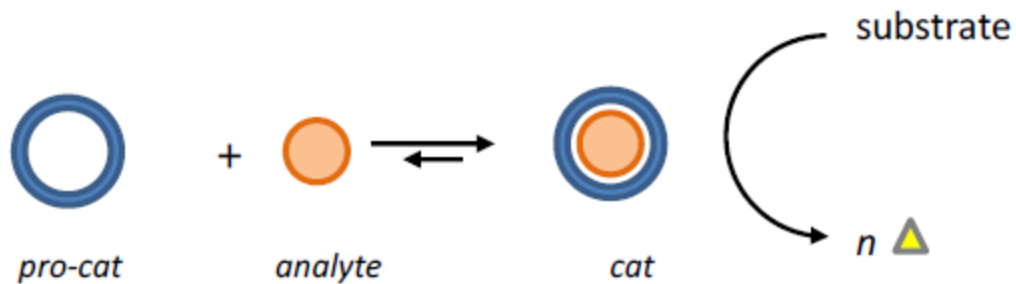
a single photon can induce a membrane hyperpolarization of 1 mV



Signal amplification in Nature - multivalency



catalysis

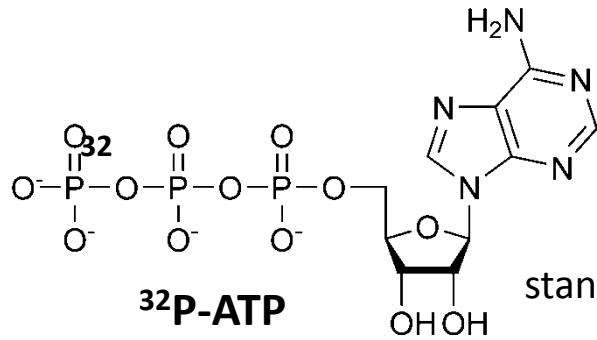
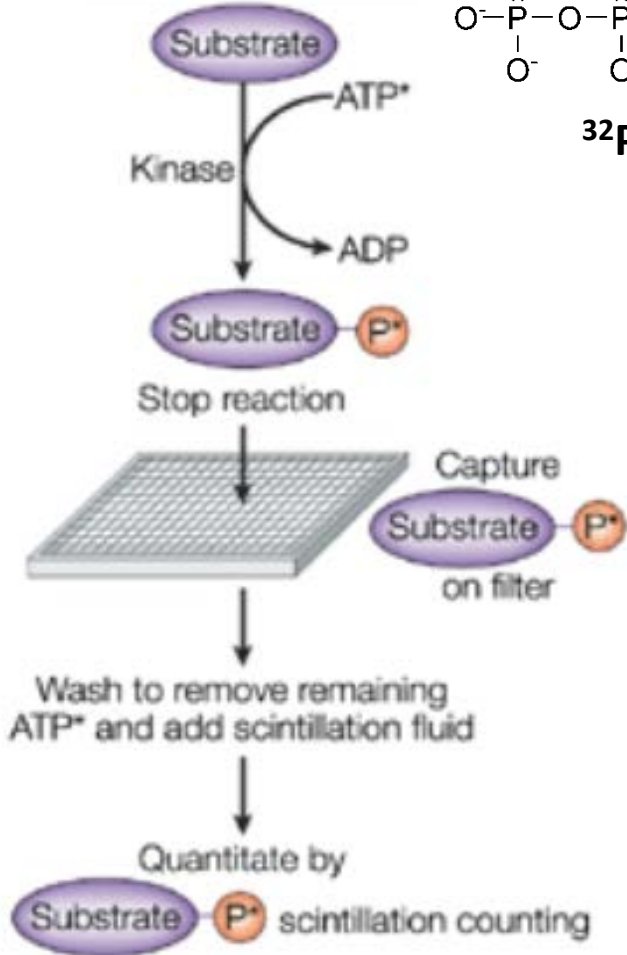


multivalency



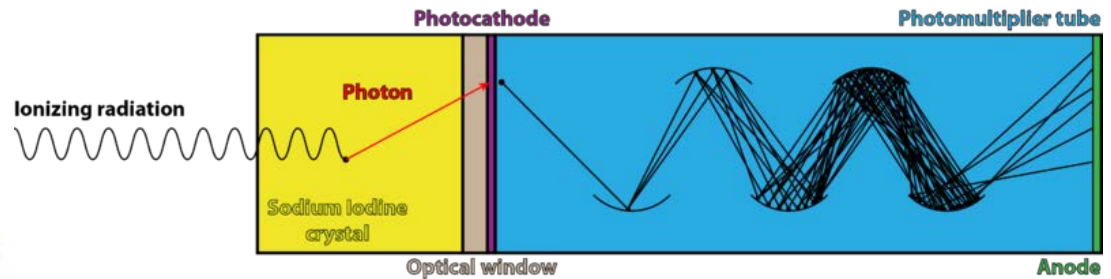


Signal amplification in assays : Radioactivity assays



quantification

standard autoradiography, phosphorimaging, or liquid scintillation counting techniques



advantage

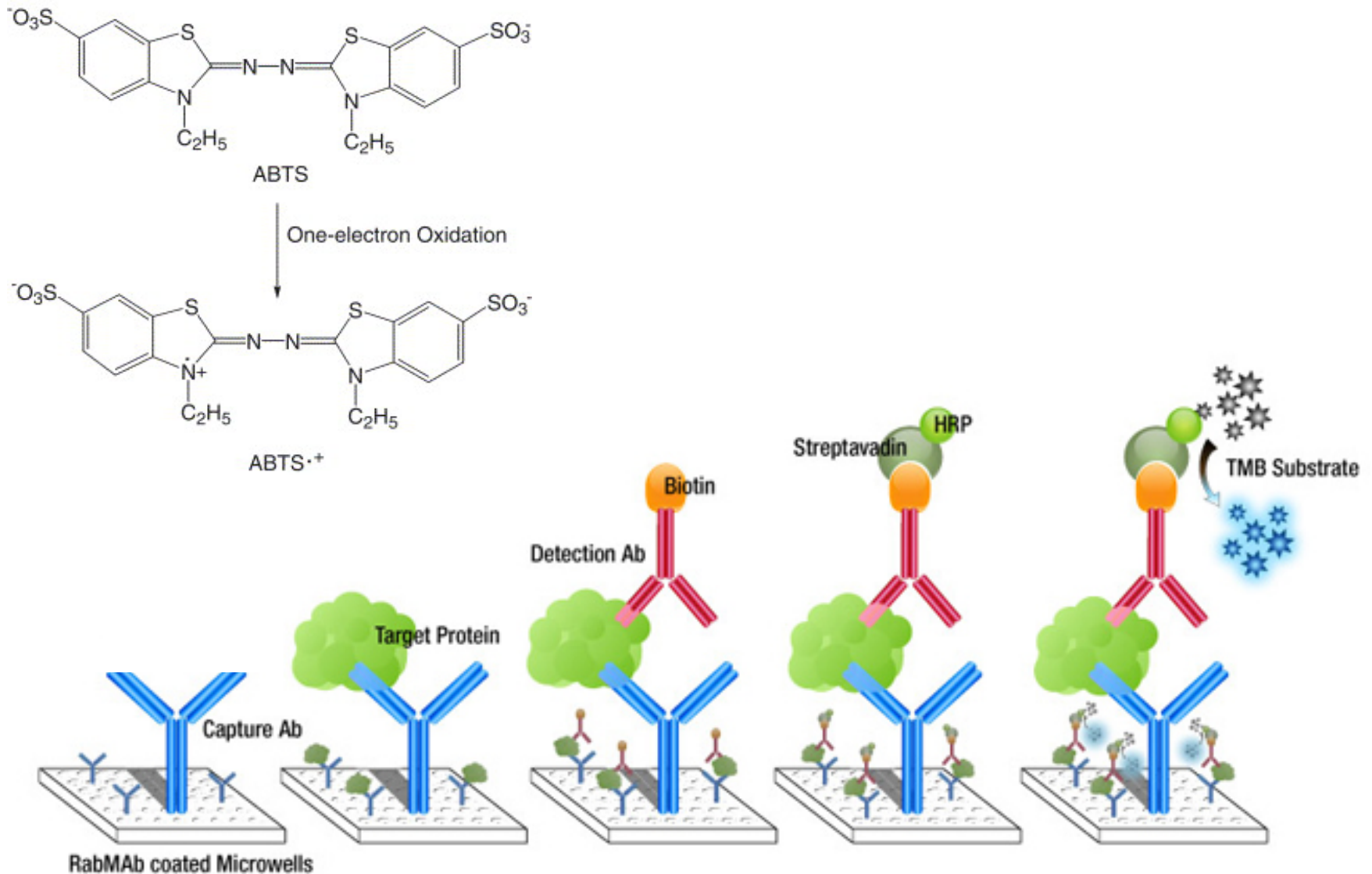
- high sensitivity
- high specificity (monitoring selective reactions)

disadvantage

- radioactivity
- no high throughput

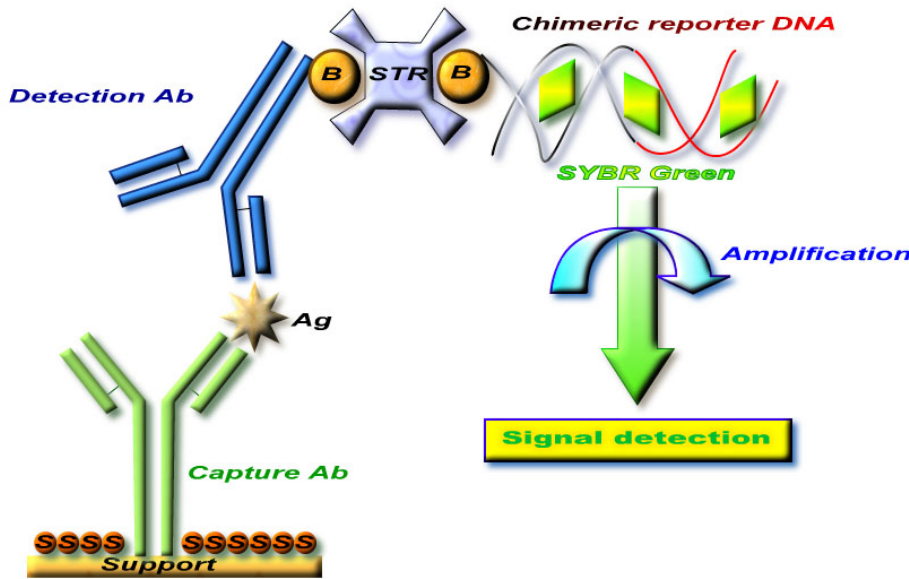


Signal amplification in assays : Enzyme-Linked ImmunoSorbent Assay (ELISA)

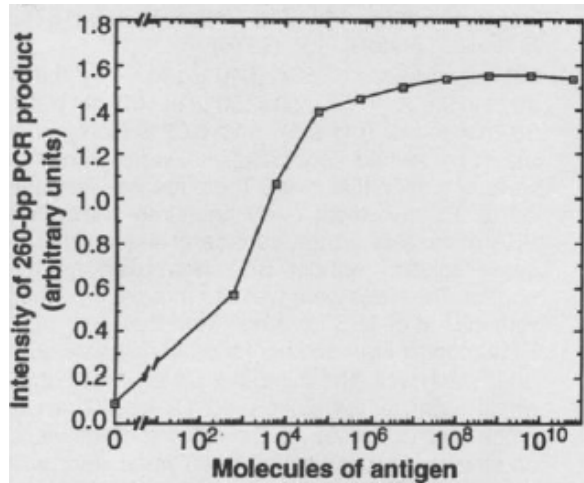




Signal amplification in assays : ImmunoPCR



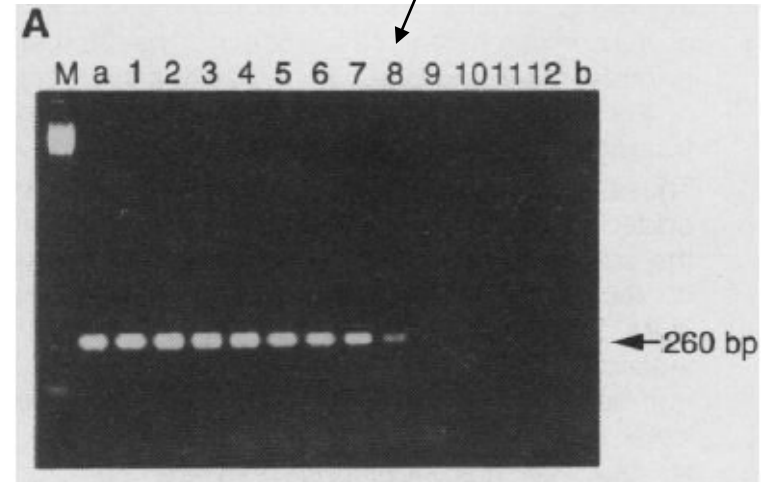
Analyte: bovine serum albumin (BSA)



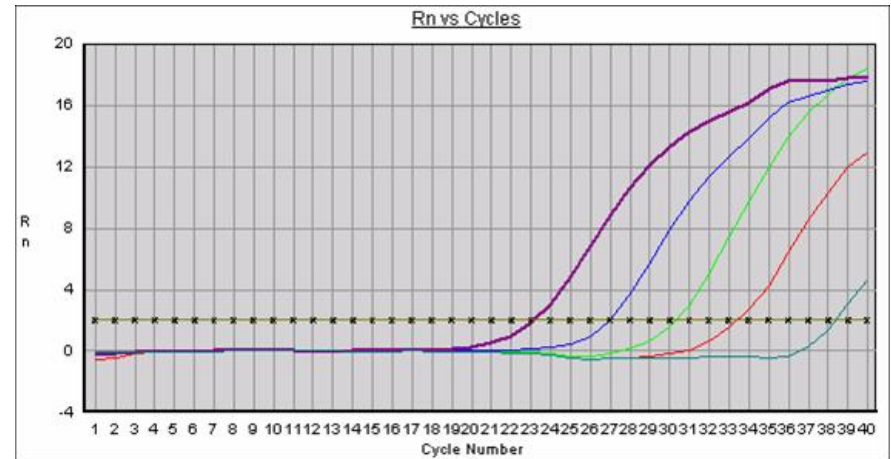
580 antigen molecules in sample
(x10⁵ increased sensitivity compared to ELISA)

9.6x10⁻²¹ mol (in 45 μl)

detection:
gel+staining (e.g. ethidium bromide)

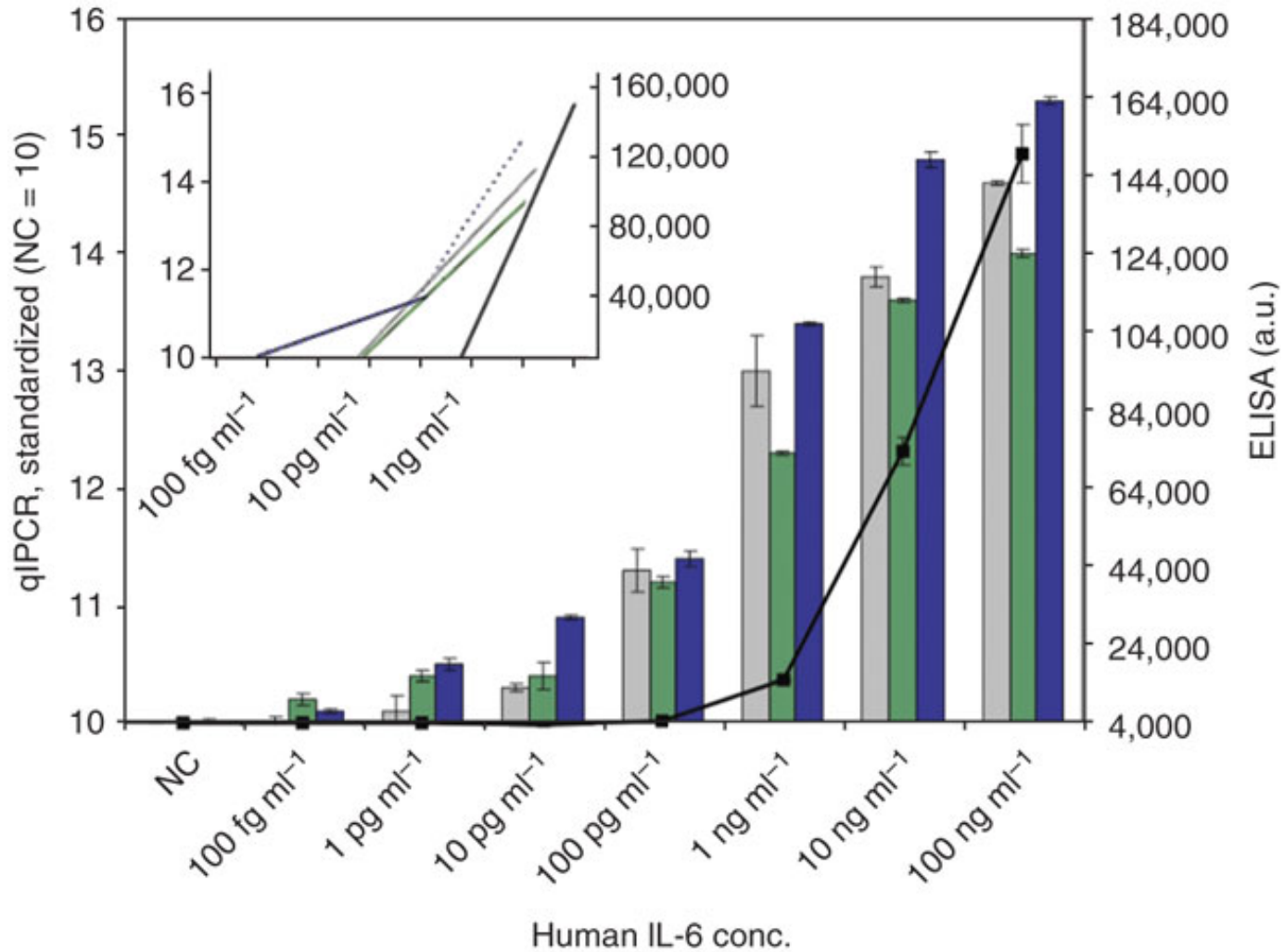


in situ by introducing fluorescent markers



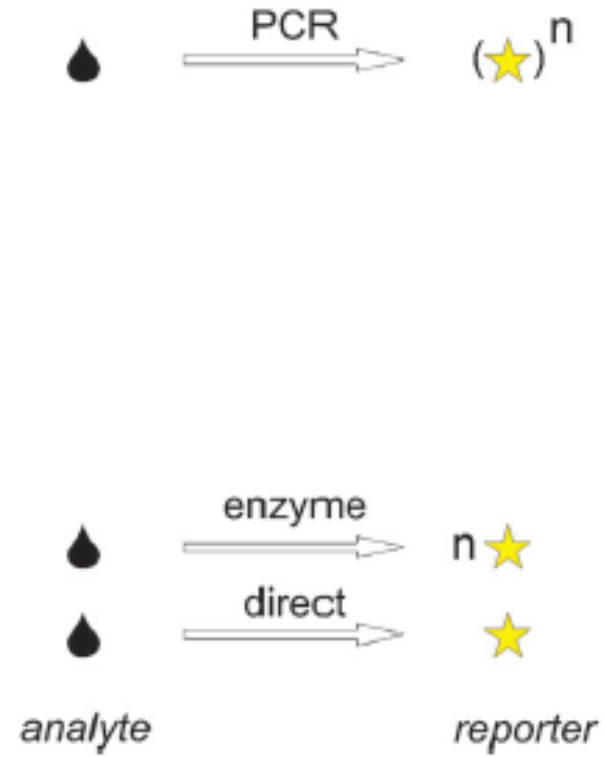
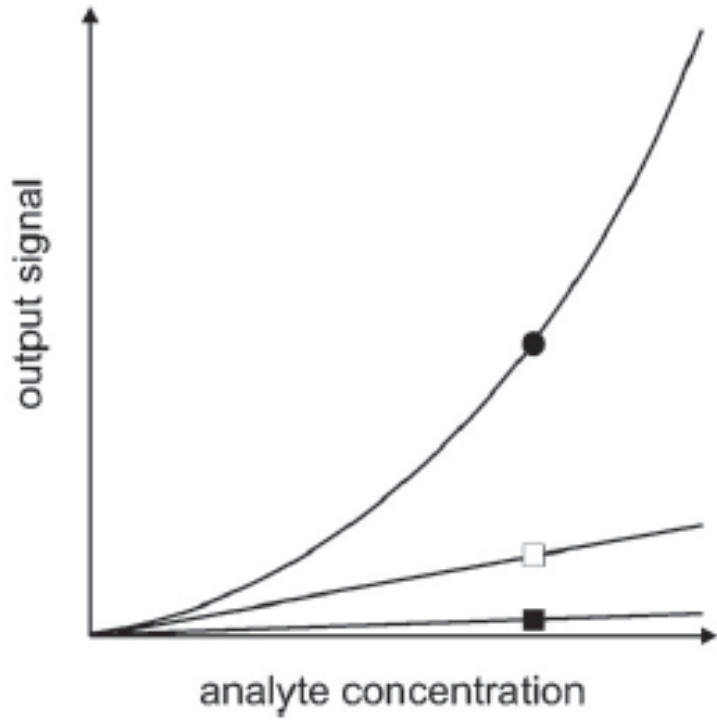


ELISA vs immunoPCR



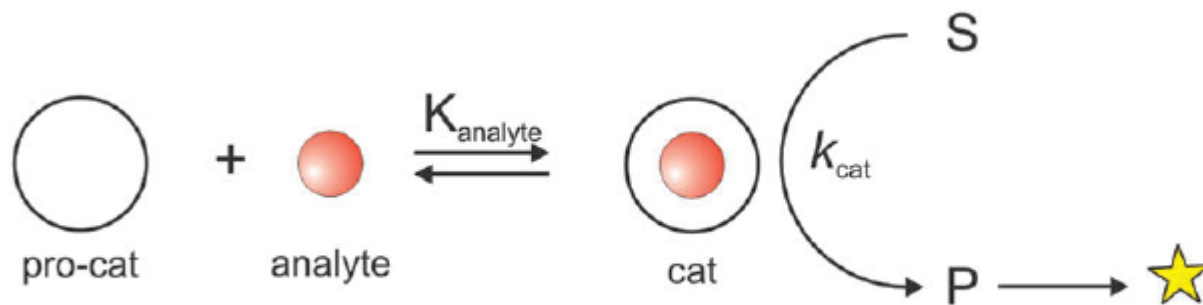


Summary





Quantification of amplification by catalyst activation



operating conditions

- high K_{analyte}
- excess pro-cat ($[\text{cat}] = [\text{A}]_0$)

catalysis

- low background rate ($k_{\text{uncat}} \ll k_{\text{cat}}$)
- detectable P
- no product inhibition

$$\frac{d[\text{P}]}{dt} = k_{\text{cat}}[\text{cat}] = k_{\text{cat}}[\text{A}]_0$$

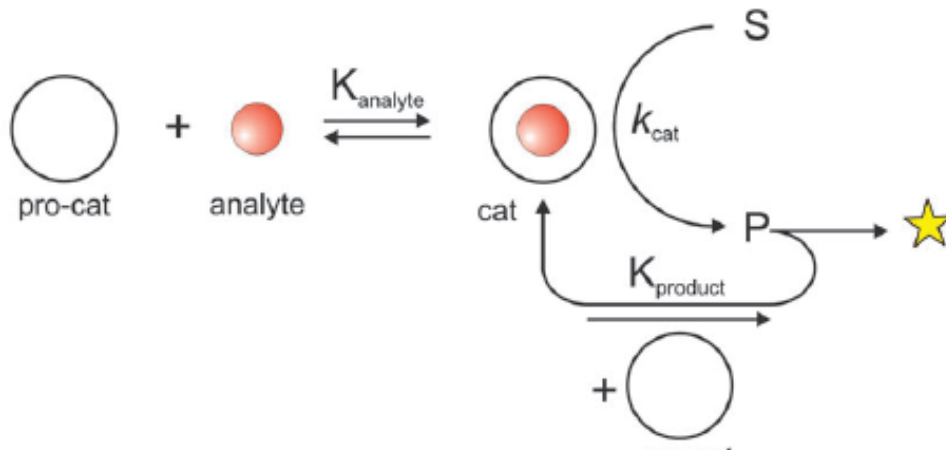


$$[\text{P}]_t = k_{\text{cat}}[\text{A}]_0 t$$

- linear relationship
- signal strength is time-dependent (high TOF)



Quantification of amplification by autocatalysis



operating conditions

- high K_{analyte} and K_{product}
- excess pro-cat ($[\text{cat}] = [A]_0$)

$$\frac{d[\text{P}]}{dt} = k_{\text{cat}}[\text{cat}] = k_{\text{cat}}[A]_0 + k_{\text{cat}}[P(t)]$$



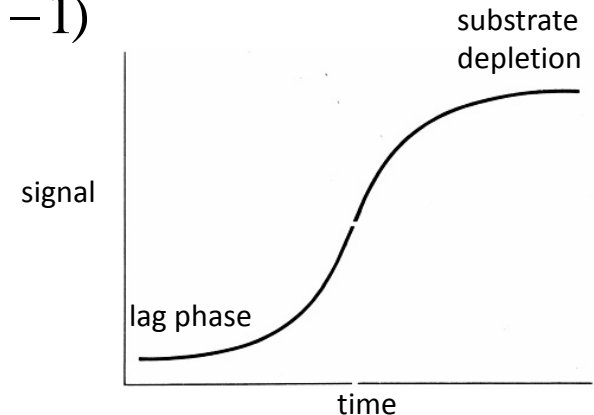
$$P(t) = P_0 e^{k_{\text{cat}} t} + A_0 (e^{k_{\text{cat}} t} - 1)$$

catalysis

in case $P_0 = 0$ $P(t) = A_0 (e^{k_{\text{cat}} t} - 1)$

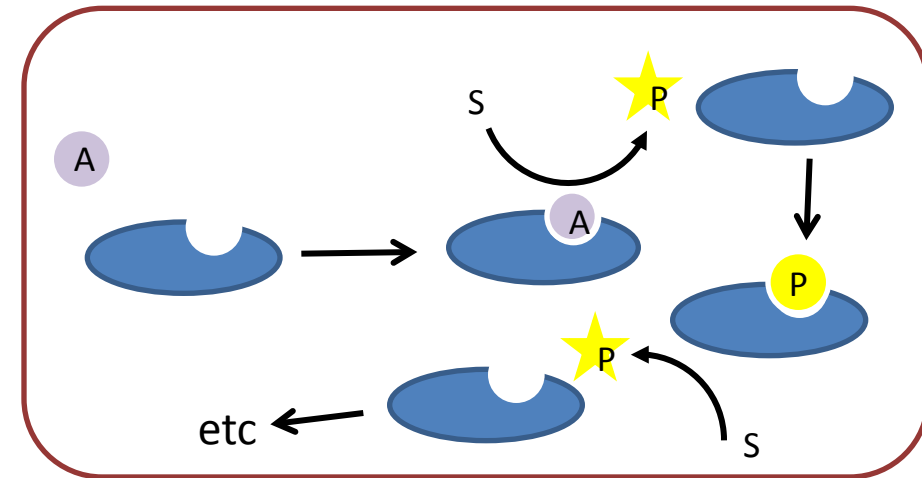
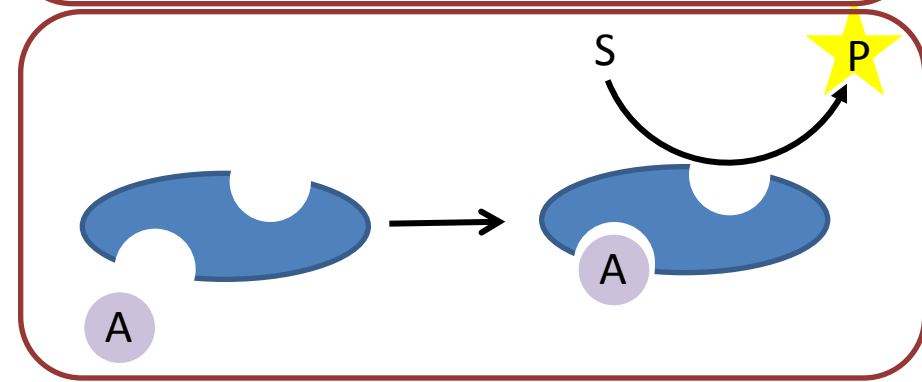
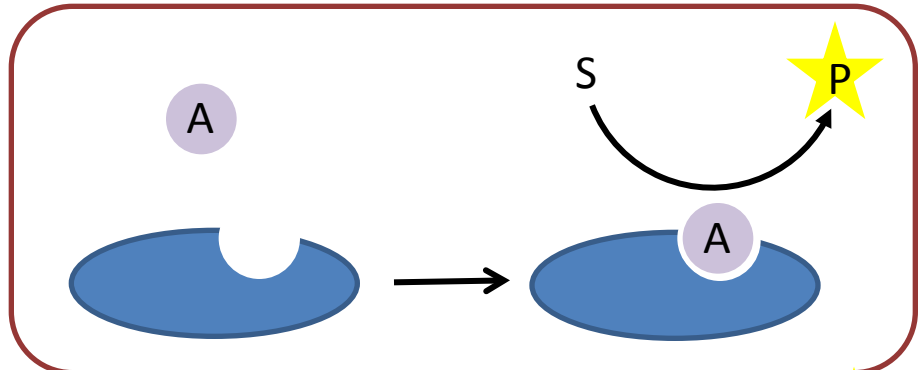
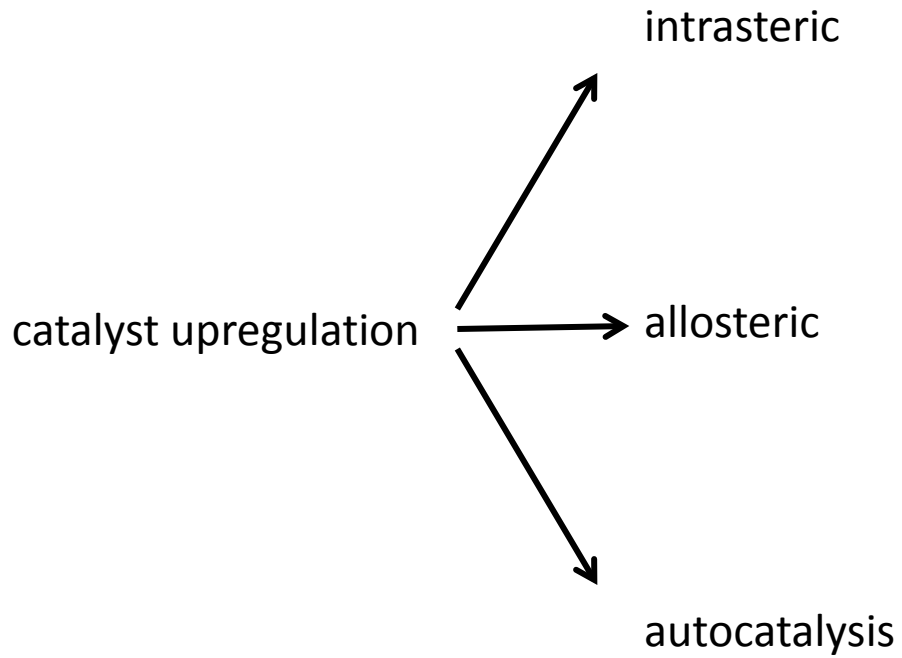
- low background rate !! ($k_{\text{uncat}} \ll k_{\text{cat}}$)
- detectable P

- exponential increase (sigmoidal curve)



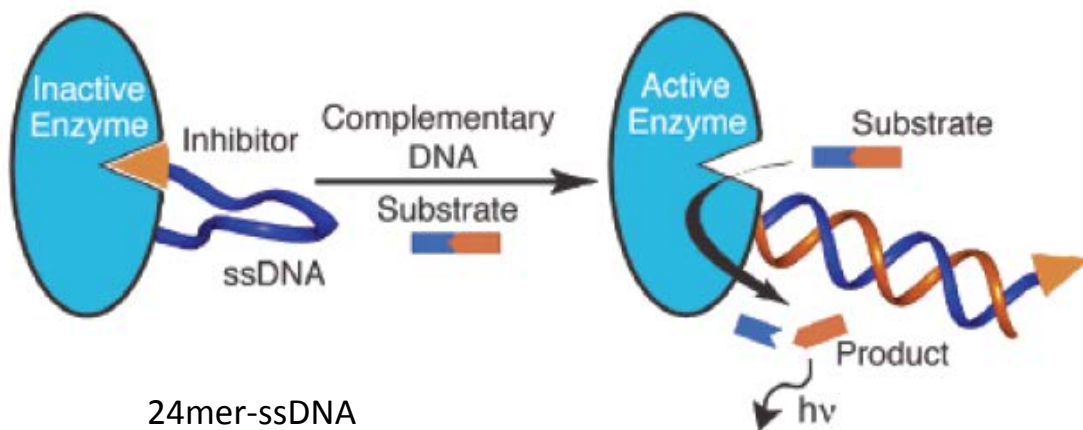


Overview examples



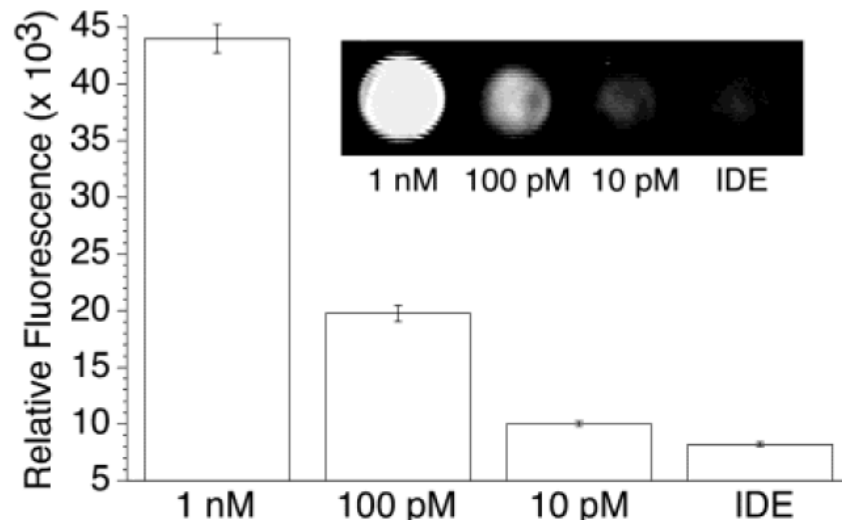
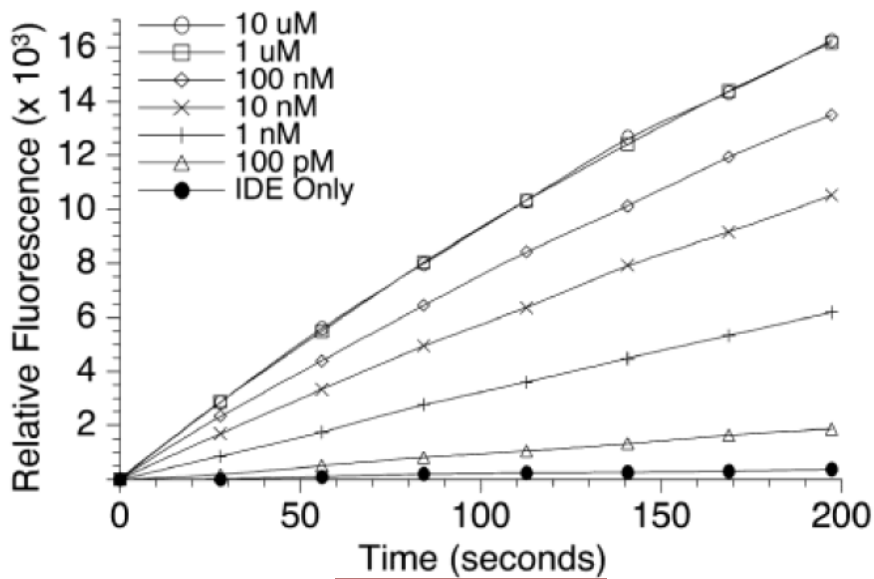


DNA detection using an engineered enzyme



IDE construct
(inhibitor-DNA-enzyme)

- Cereus neutral protease (CNP)
- compatible with DNA-tether
 - high activity ($k_{cat} = 165 \text{ s}^{-1}$)

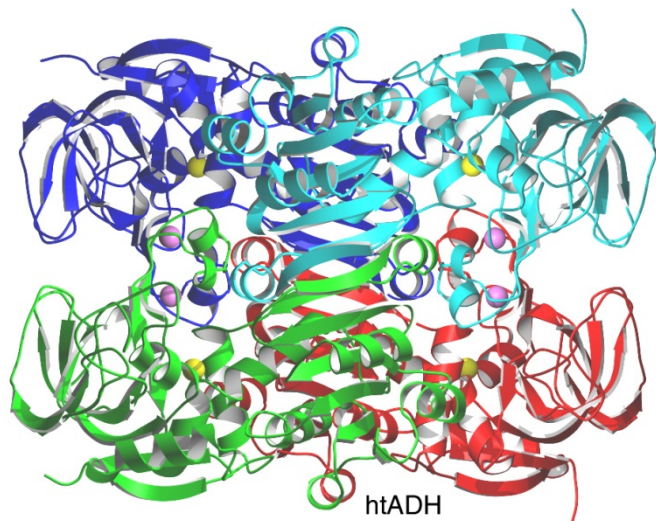


readout after 80 minutes

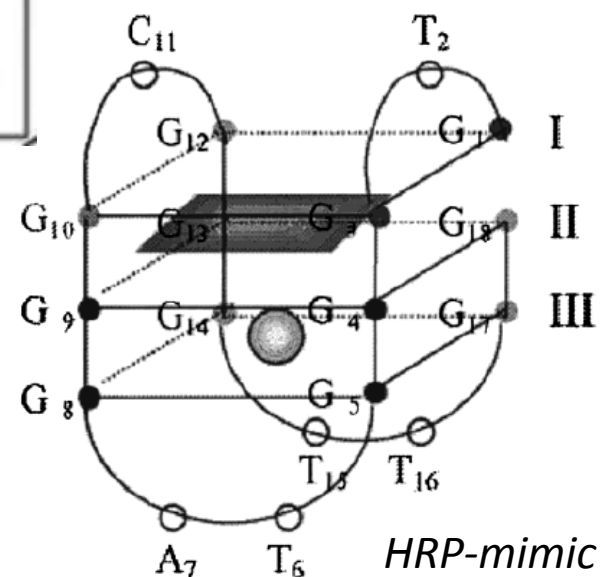
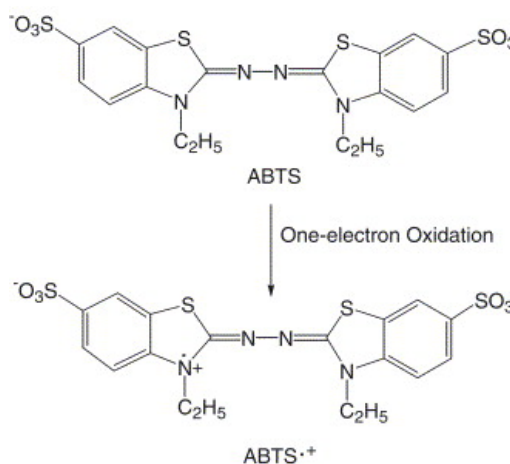
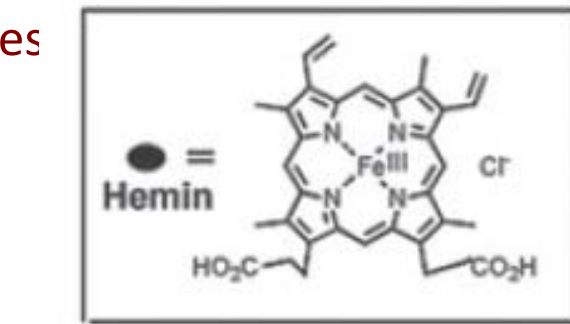
Issues: enzyme modification, synthesis



Enzymes versus DNAzymes



enzymes



DNAzymes

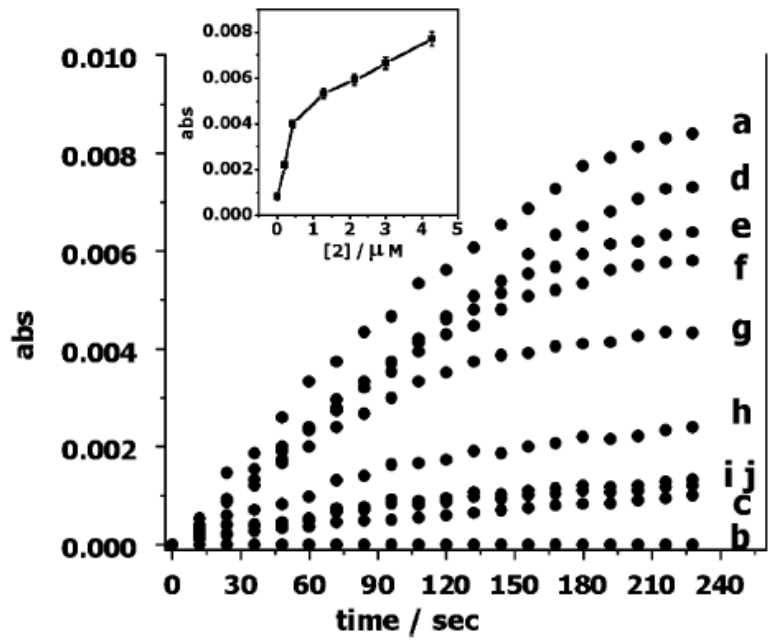
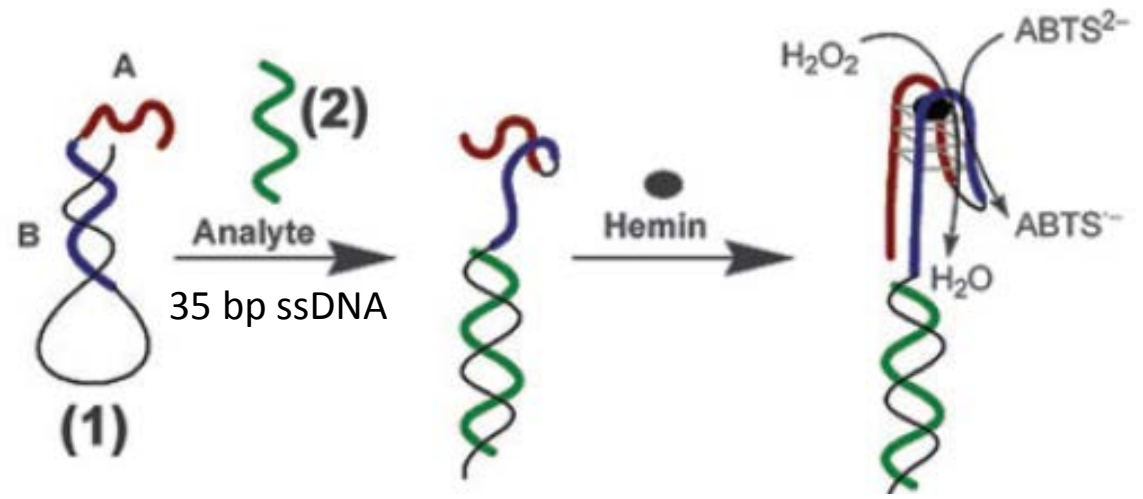
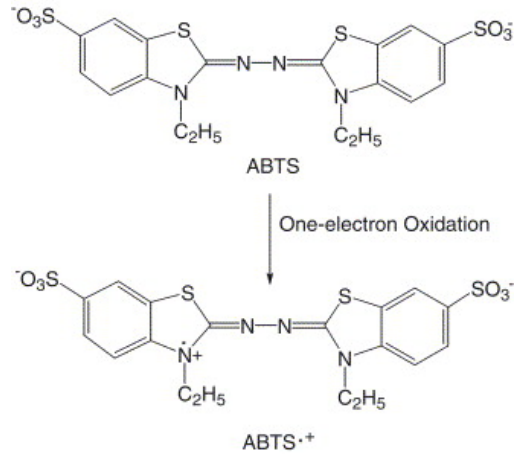
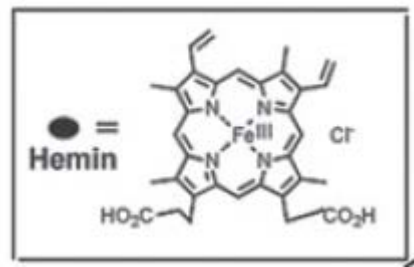
($\epsilon = 3.6 \times 10^4 \text{ M}^{-1} \text{ cm}^{-1}$) (DNA with catalytic activity)

- high activity
- selective conditions
- isolation/preparation in small quantities

- nonnatural oligonucleotide-based catalysts
- higher chemical stability
- easy conjugation to other molecules (biotin)
- preparation in large quantities using PCR



DNA detection using a DNAzyme



4.3 μM

↓

0.2 μM

mismatch DNA (single bp mismatch)
control (no analyte)
control (no beacon)

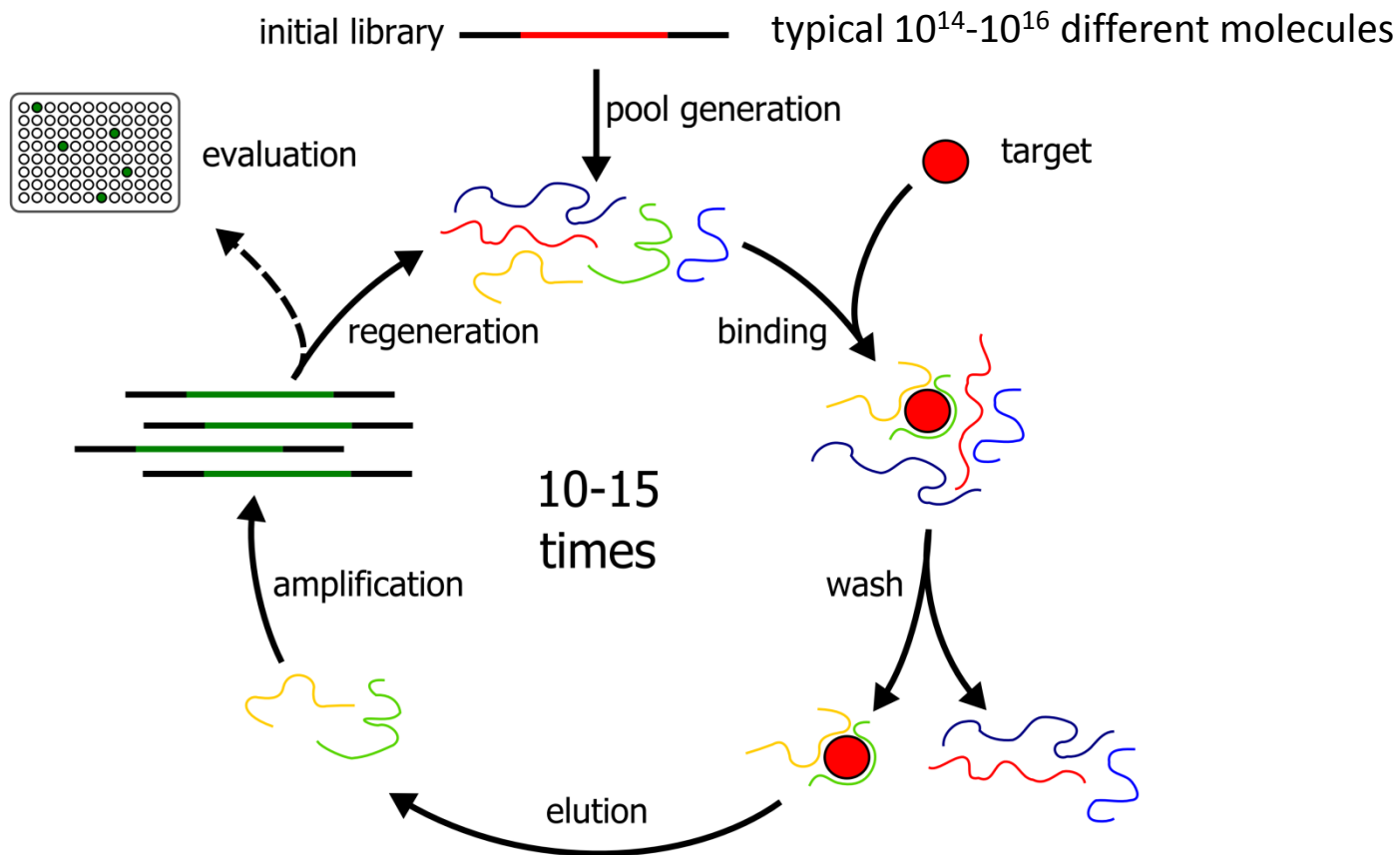
- mM
- μM
- nM
- pM
- fM
- aM
- zM
- yM



Detection of small molecules using aptazymes

Aptamers are nucleic acid molecules (DNA/RNA) with specific binding functionality. In the presence of targets, aptamers adopt a complementary 3D conformation which can preferentially bind the targets

Aptazymes are rational combination of DNAzymes/ribozymes and aptamers.



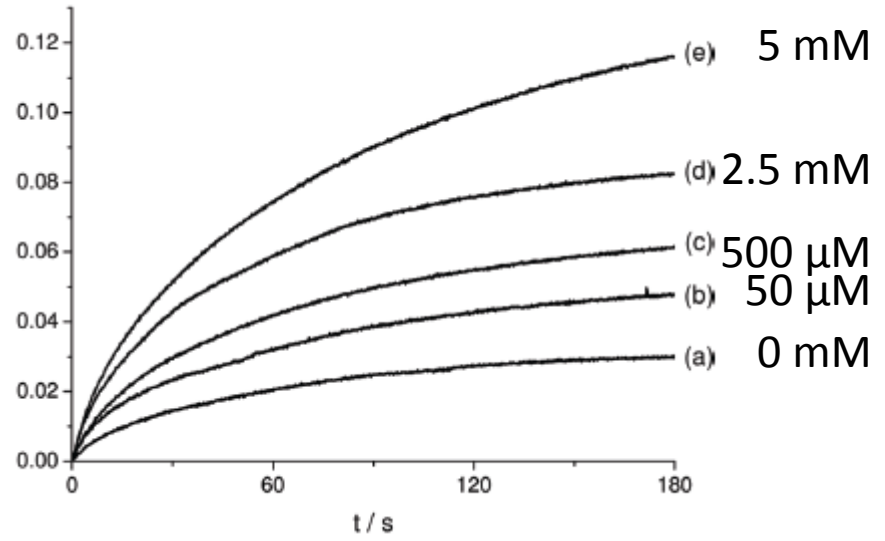
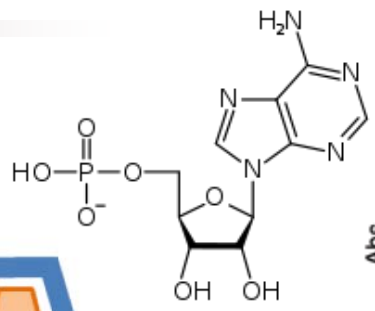
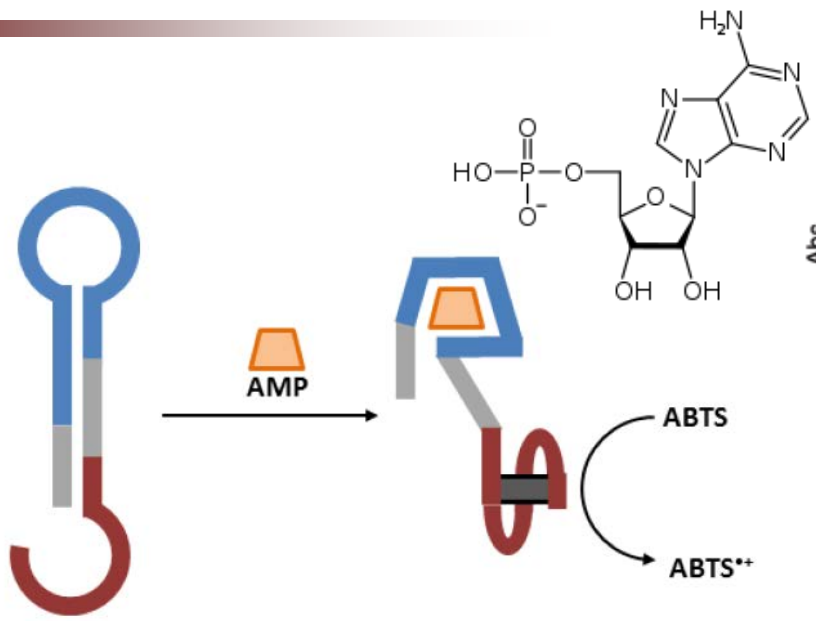
SELEX

Systematic Evolution of Ligands by EXponential Enrichment



Detection of AMP using an aptazyme

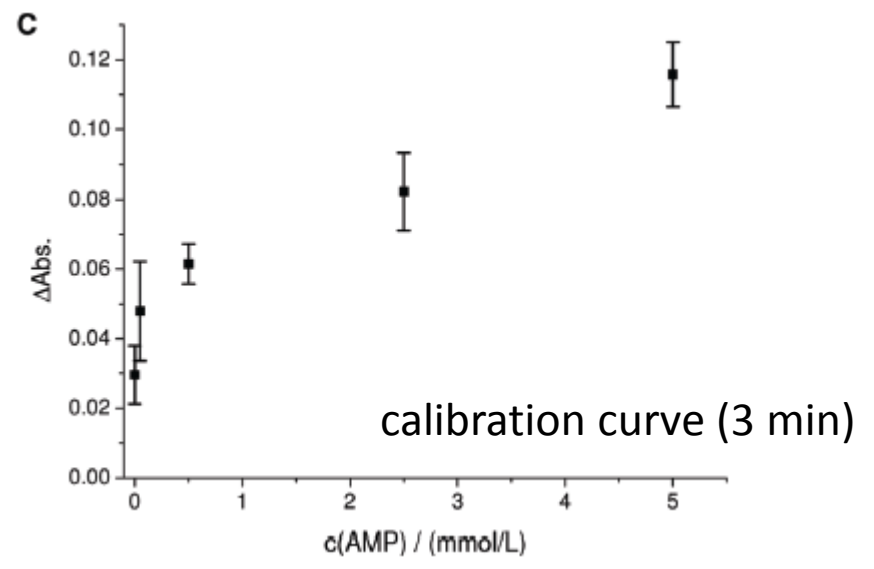
- mM
- μM
- nM
- pM
- fM
- aM
- zM
- yM



Reason for low sensitivity

AMP-aptamer complex formation competes with duplex formation.

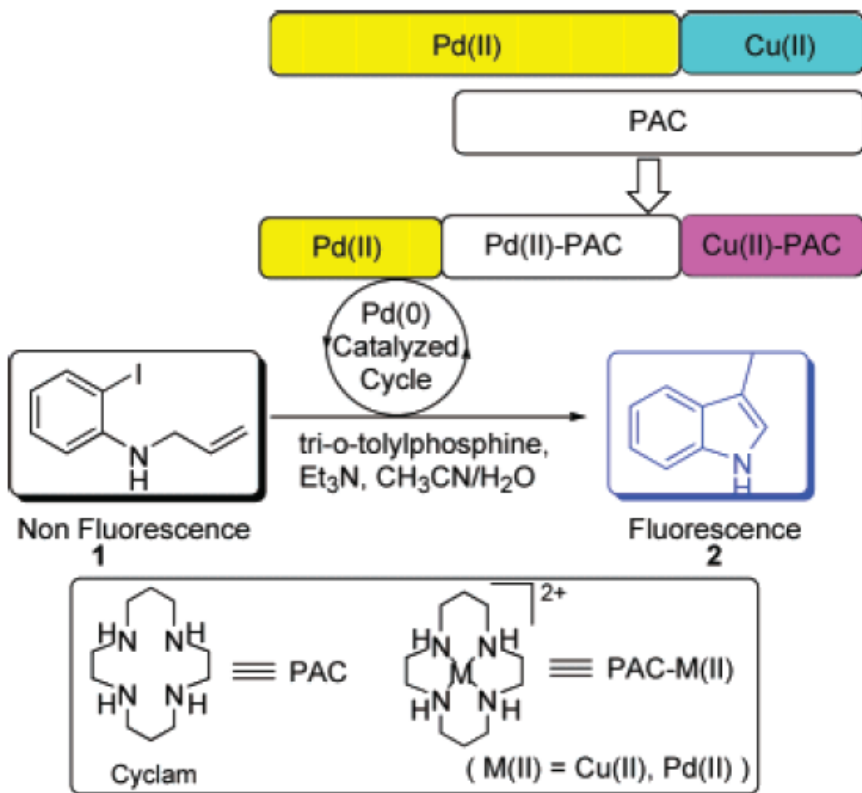
Reported: stability of the stem duplex ($\Delta G^\circ = -9.62 \text{ kcal}\cdot\text{mol}^{-1}$) was adjusted to that of the aptamer-AMP complex ($K_{\text{eq}} = 1.67 \times 10^5 \text{ M}^{-1}$; , $\Delta G^\circ = 7.1 \text{ kcal}\cdot\text{mol}^{-1}$).





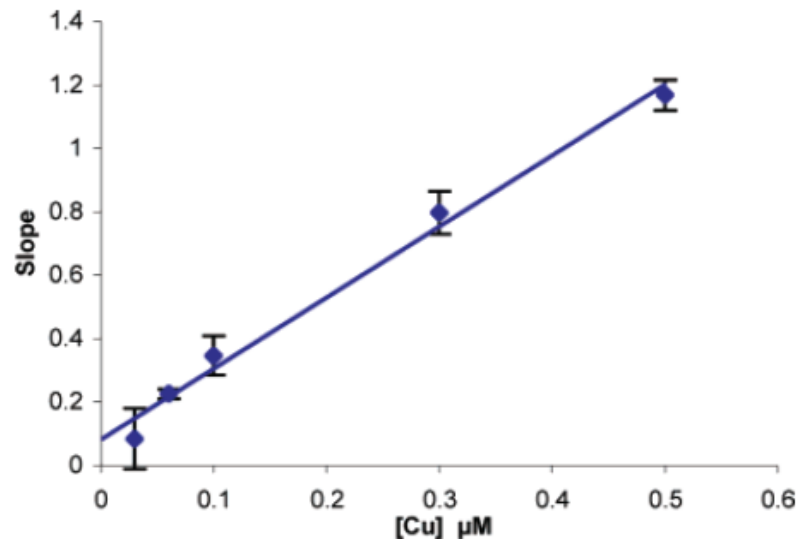
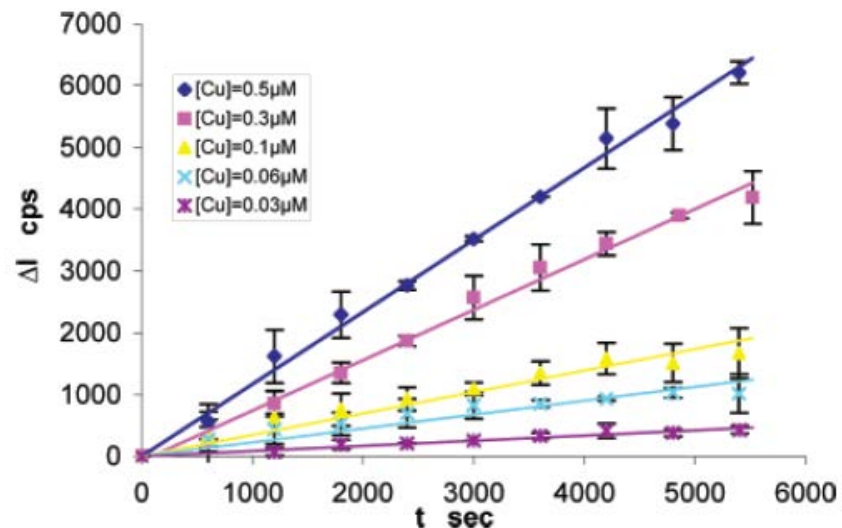
Metal detection using a synthetic system

$$K_{\text{Cu(II)} \cdot \text{PAC}} \gg K_{\text{Pd(II)} \cdot \text{PAC}}$$



30 nM analyte requires 90 minutes

The higher the binding affinity (series Cu(II), Ni(II), Co(II), and Cd(II)), the greater the initial rate observed (**selectivity**).



mM

μM

nM

pM

fM

aM

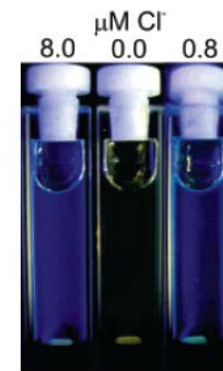
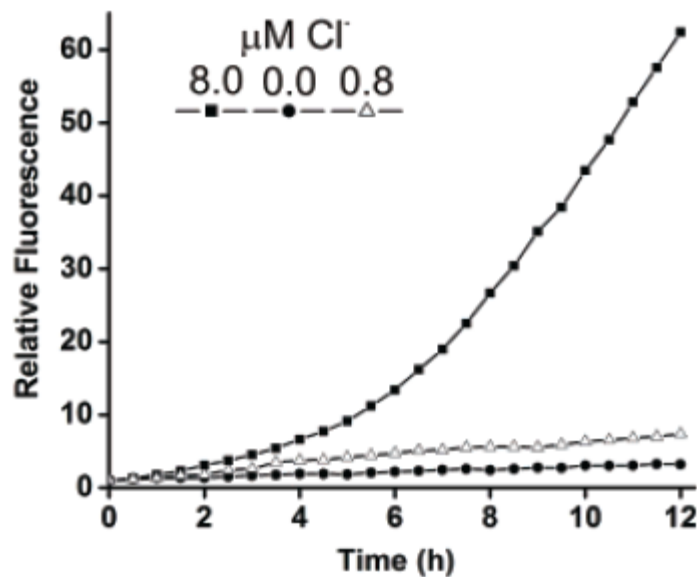
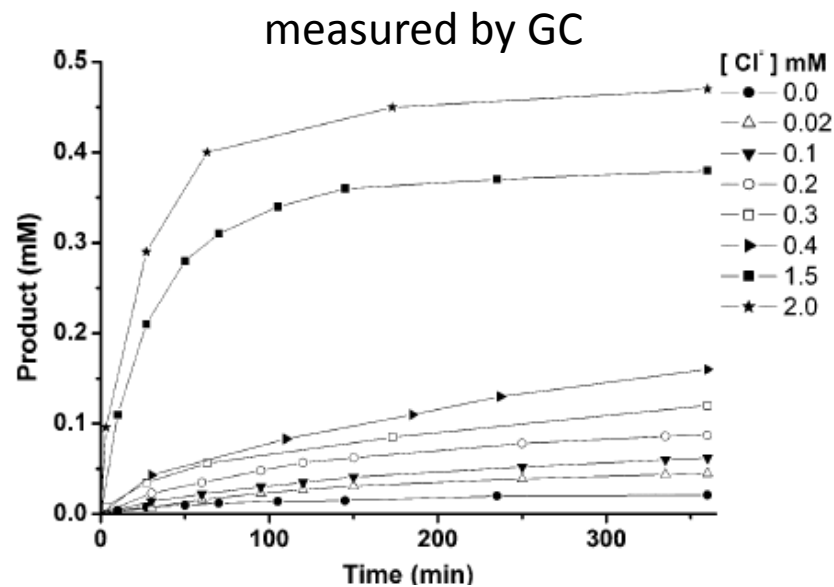
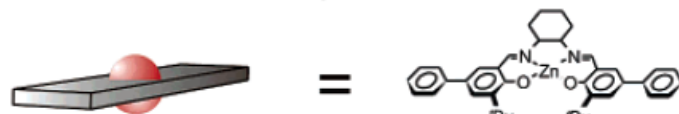
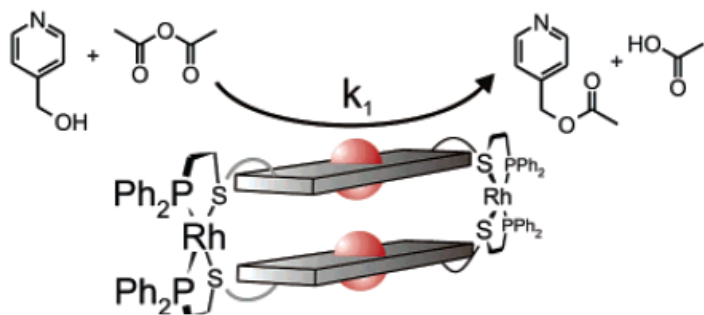
zM

yM



Small molecule detection using allosteric control

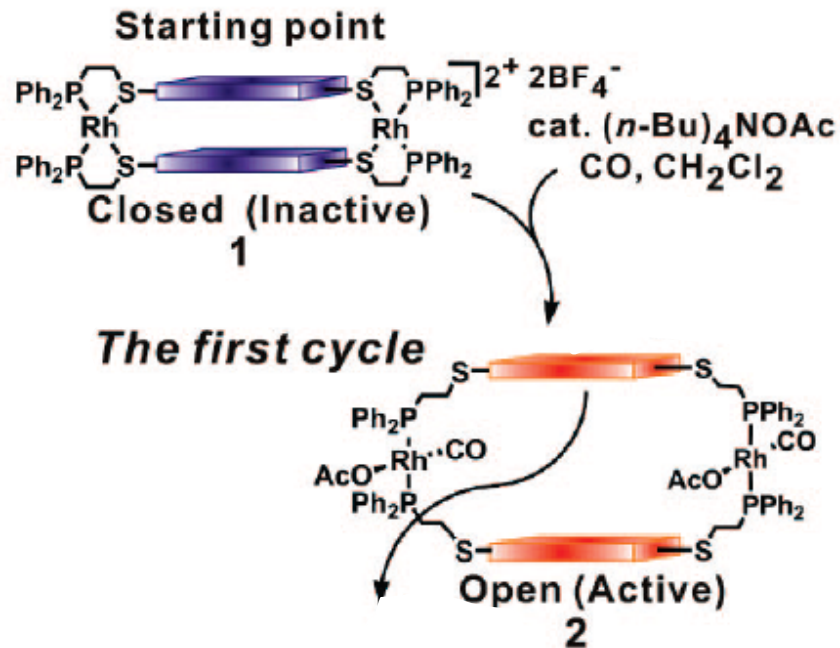
- mM
- μ M
- nM
- pM
- fM
- aM
- zM
- yM



6 hours



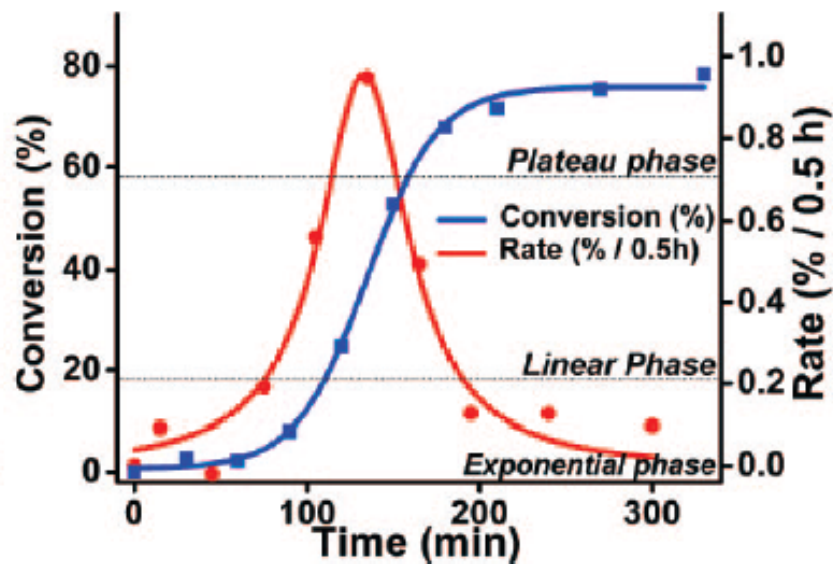
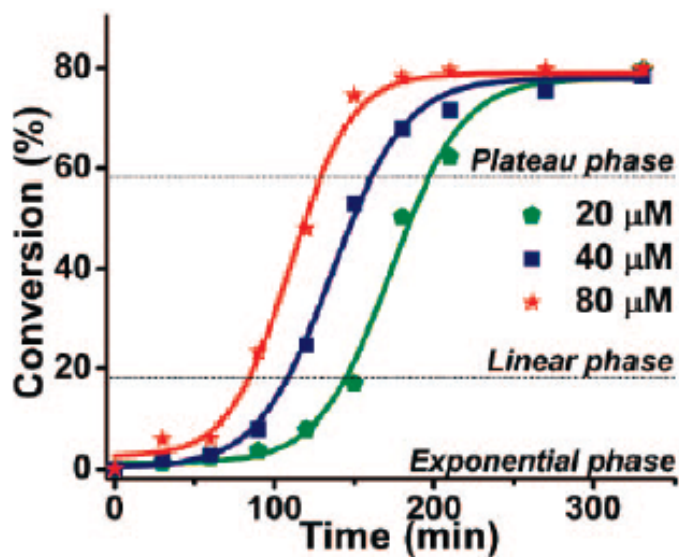
Cascade reactions





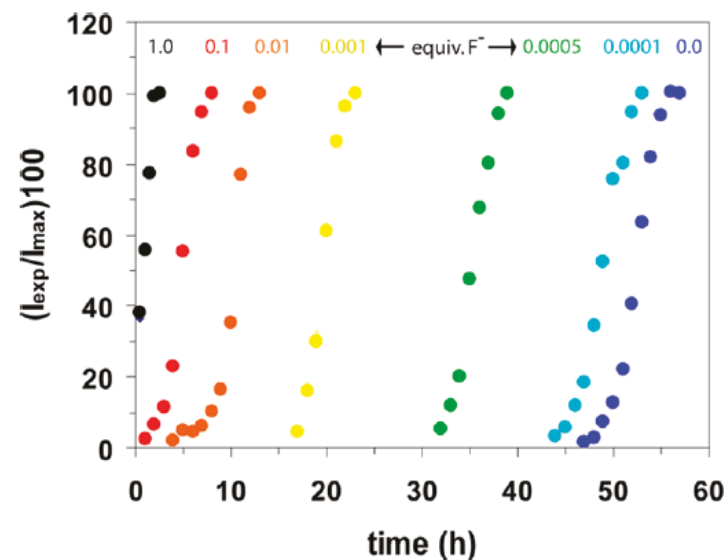
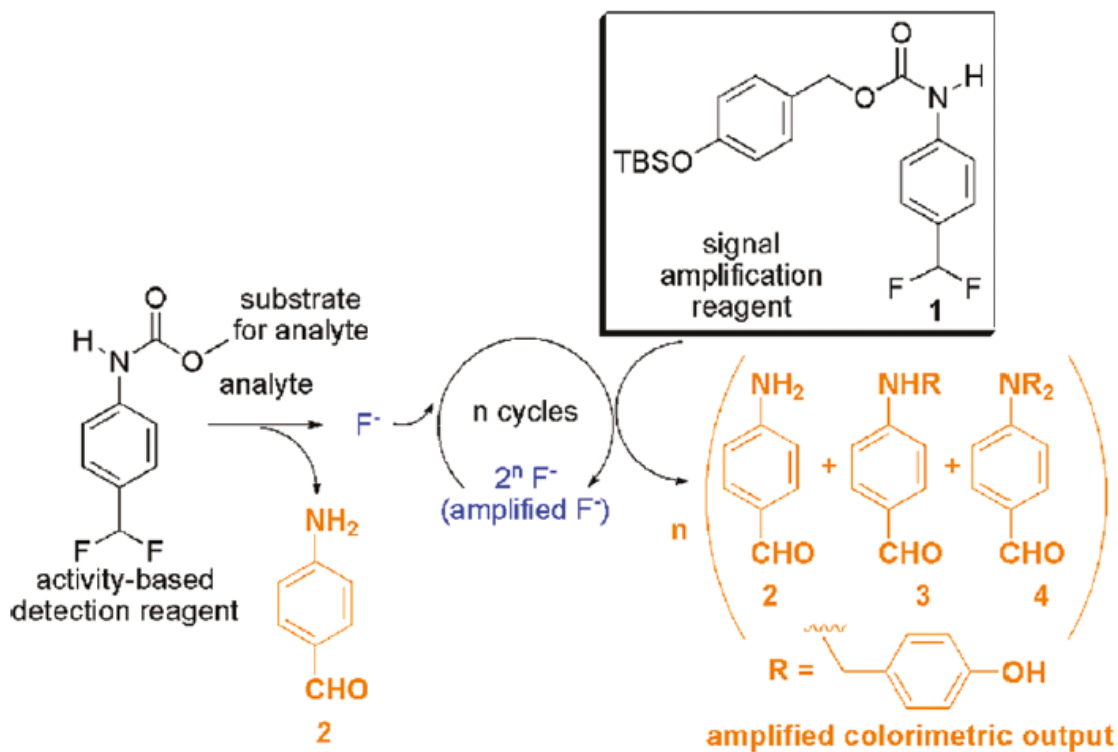
PCR-like amplification

- mM
- μ M
- nM
- pM
- fM
- aM
- zM
- yM





A Two-Component Small Molecule System for the detection of Pd(II)



response to F^-

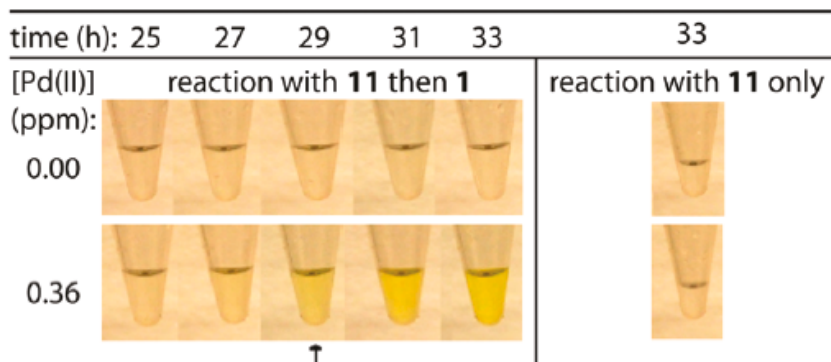
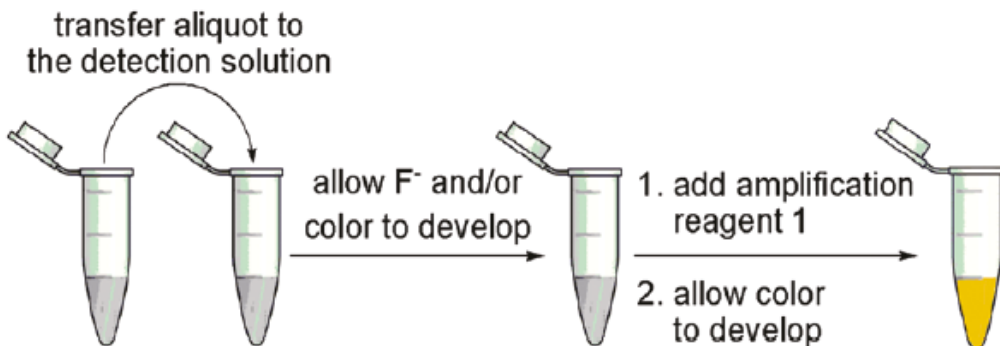
$$\alpha = \frac{(I_{amp} - I_{background})}{I_{initial}}$$

α reaches 2168 (35 for 0.0005 equiv F^-)
corresponding to 11 amplification cycles

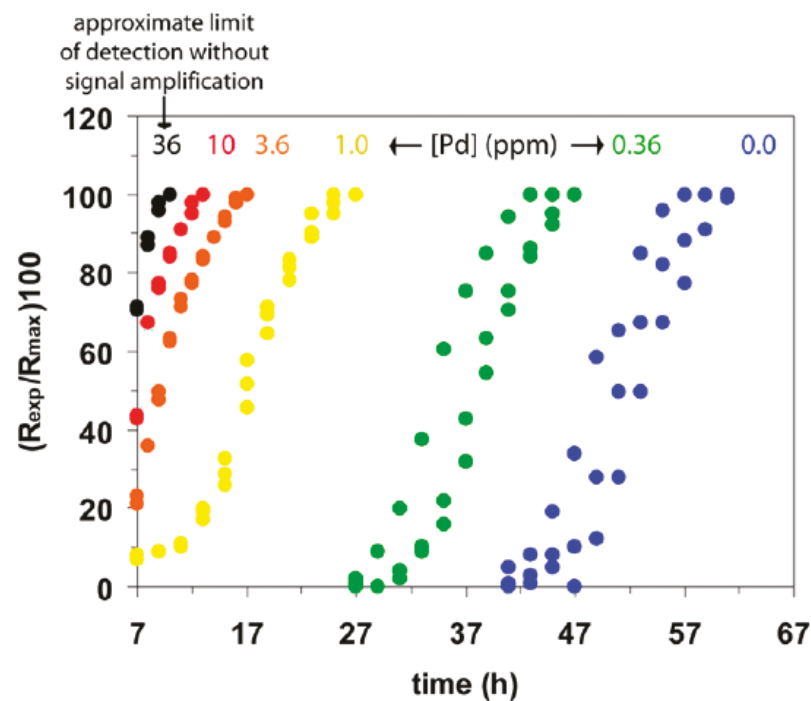


Assay development

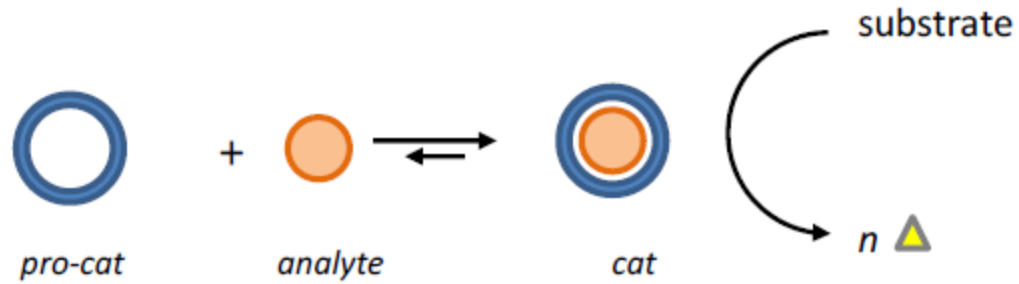
mM
 μM
 nM
 pM
 fM
 aM
 zM
 yM



20% of the colorimetric signal occurs at ~29 h

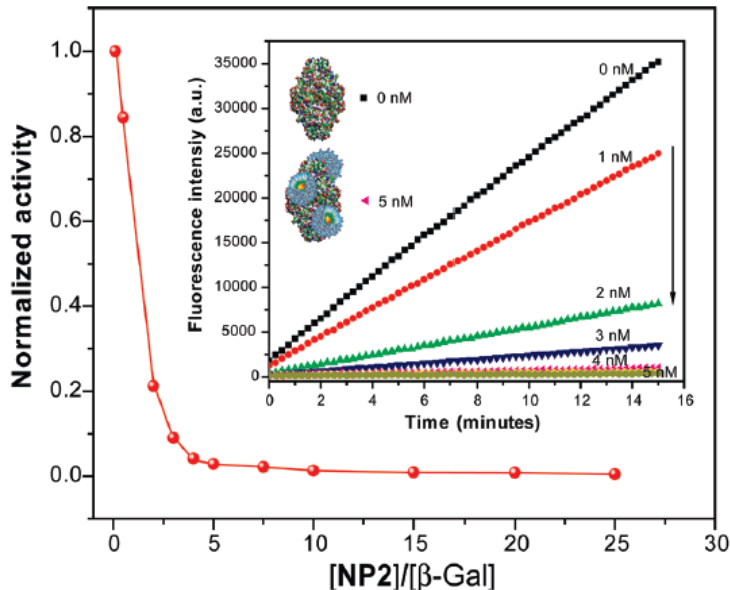
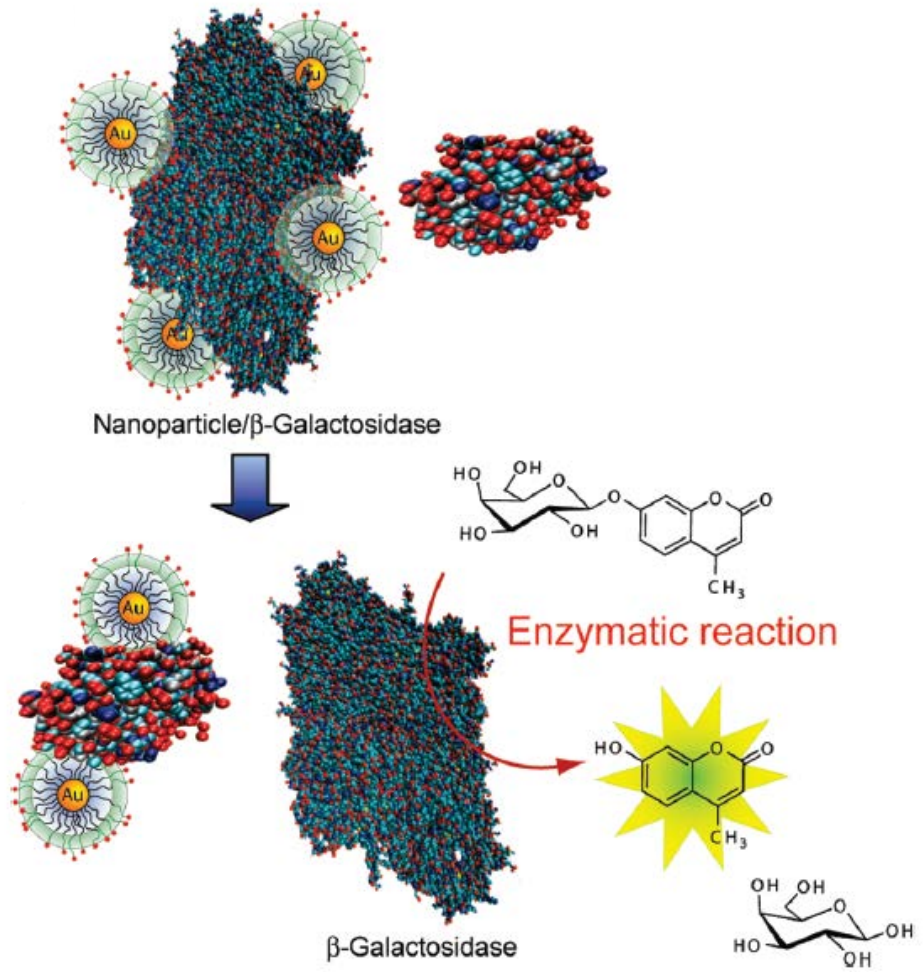
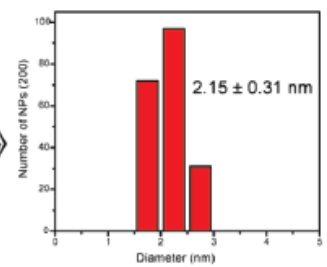
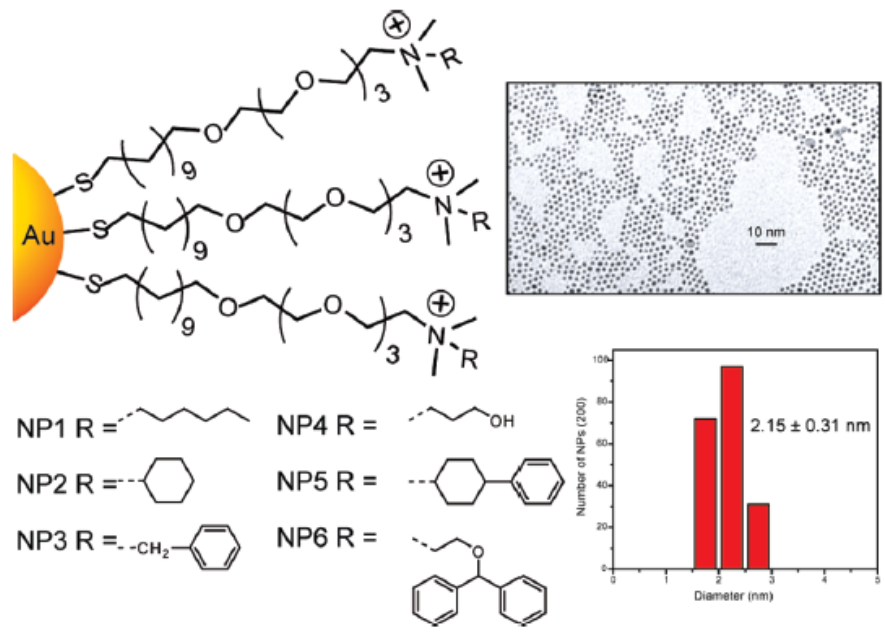


catalysis



from the sensing of small molecules
to
the sensing of proteins

Nanoparticle based detection of proteins



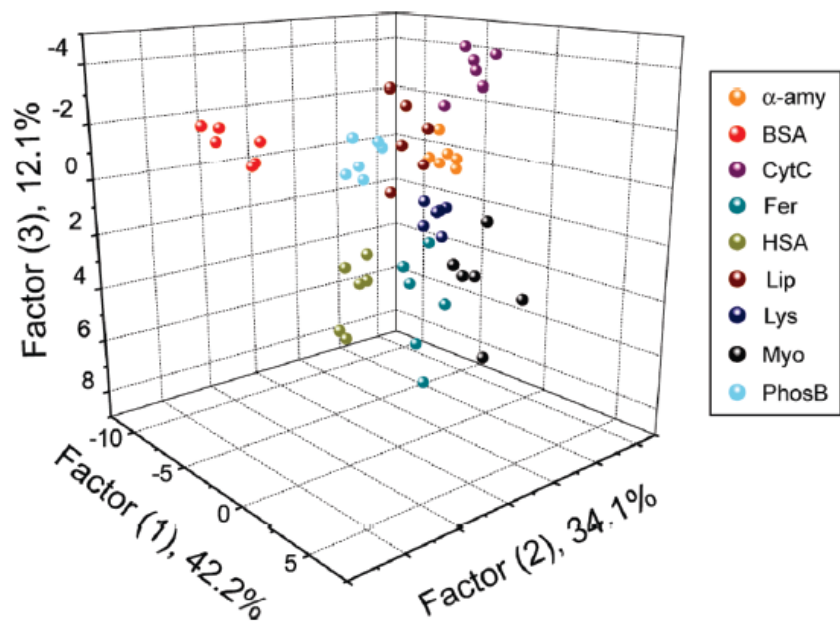
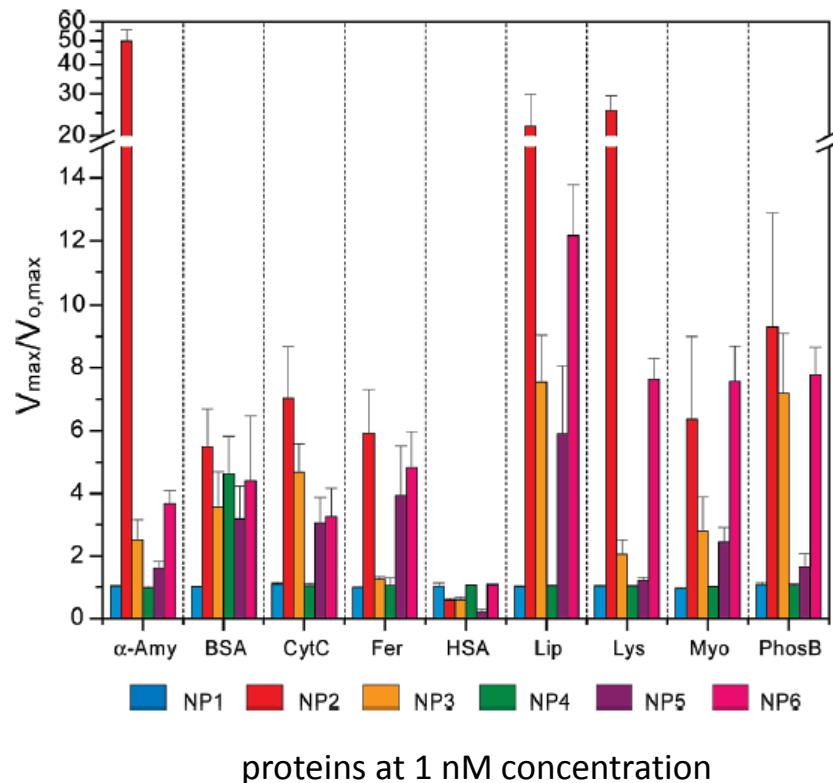


Sensing of proteins

Table 1. Physical Properties of the Proteins Used As Sensing Targets in Phosphate Buffer Solution at pH 7.4^{18a}

protein ^a	M _w (kDa)	pI
<i>α-amylase (α-Am)</i>	50.0	5.0
bovine serum albumin (BSA)	66.3	4.8
cytochrome <i>c</i> (CytC)	12.3	10.7
<i>ferritin (Fer)</i>	750.0	4.5
<i>human serum albumin (HSA)</i>	69.4	5.2
<i>lipase (Lip)</i>	58.0	5.6
<i>lysozyme (Lys)</i>	14.4	11.0
<i>myoglobin (Myo)</i>	17.0	7.2
<i>alkaline phosphatase (PhosB)</i>	140.0	5.7

^a Proteins in *italics* are commonly found in human urine.



mM

μM

nM

pM

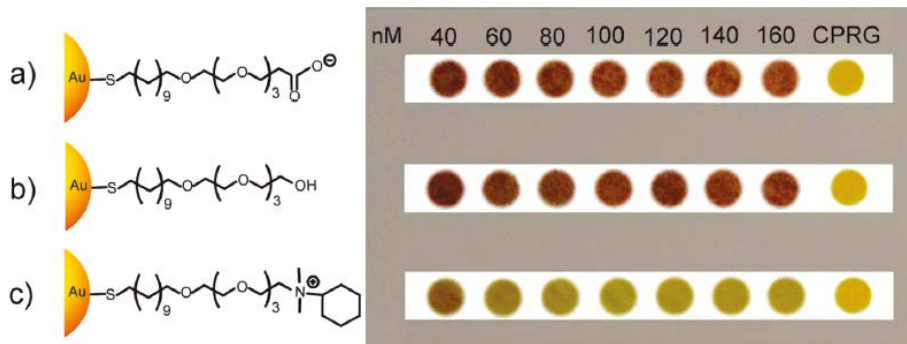
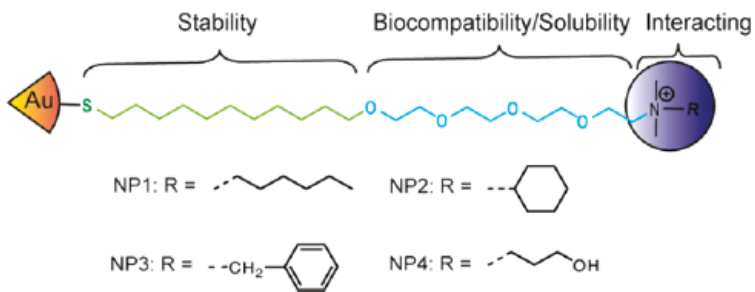
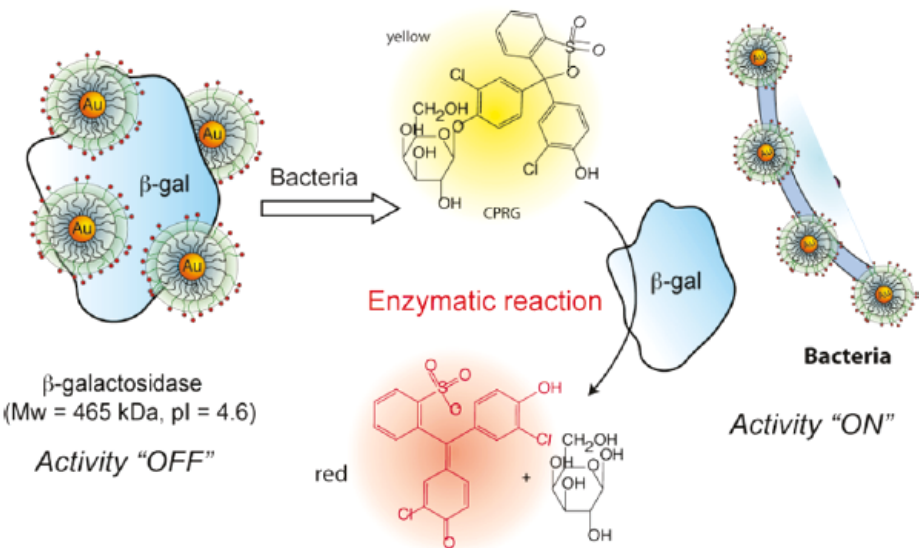
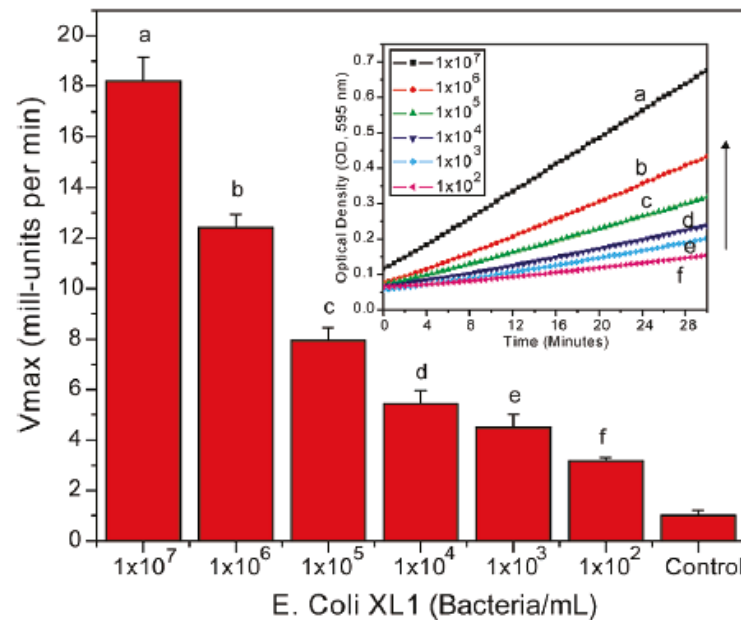
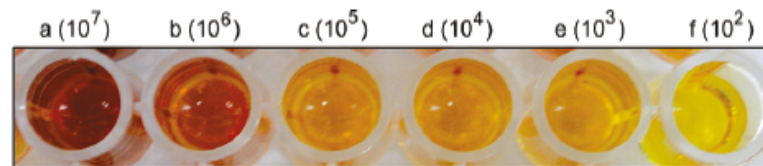
fM

aM

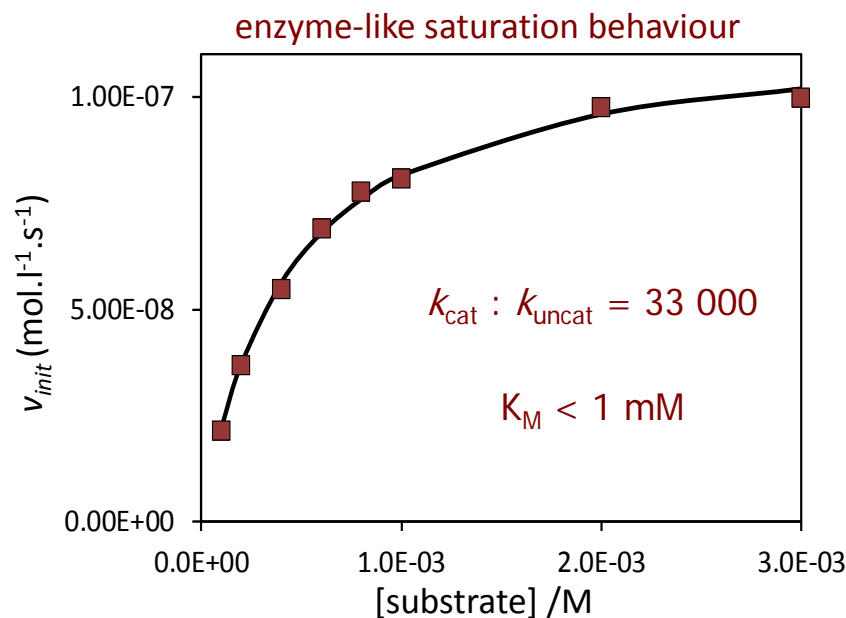
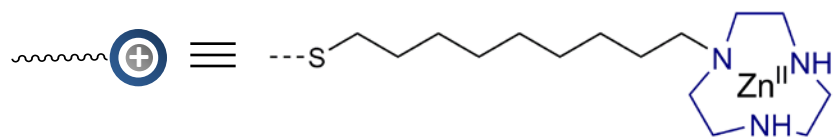
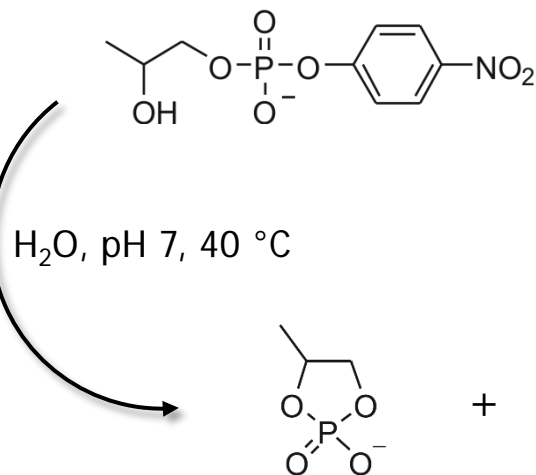
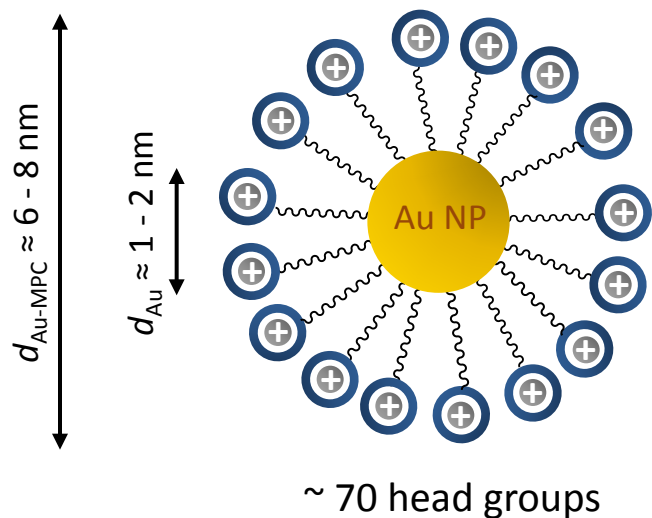
zM

yM

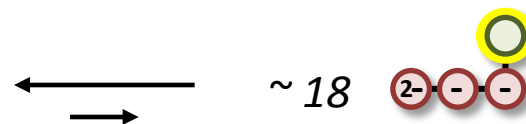
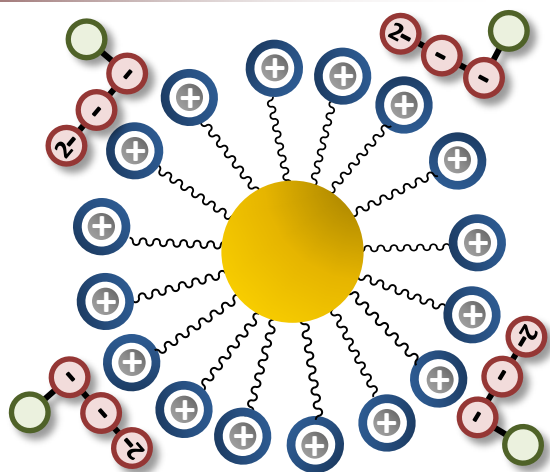
Sensing of bacteria



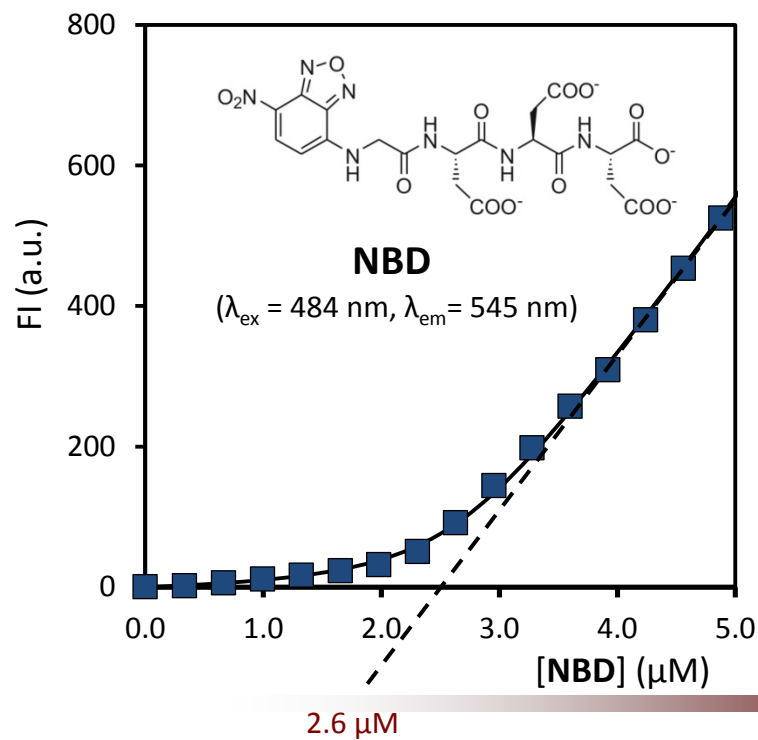
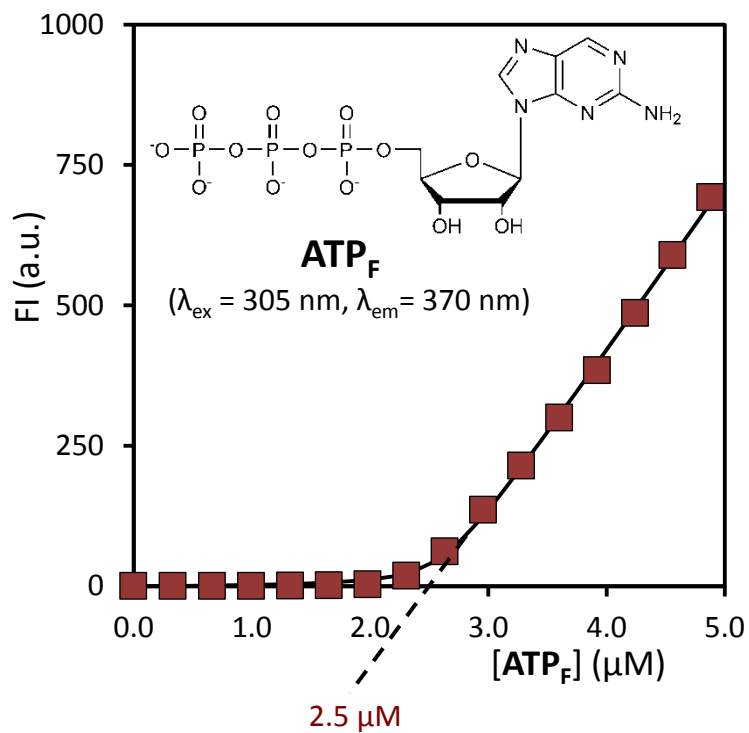
Catalytic nanoparticles for sensing



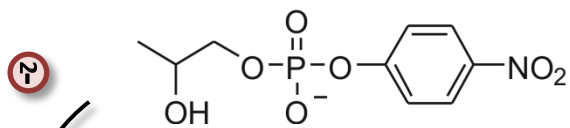
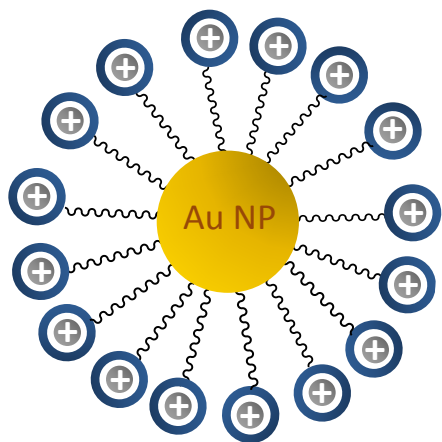
Monolayer protected Au nanoparticles: cationic surface



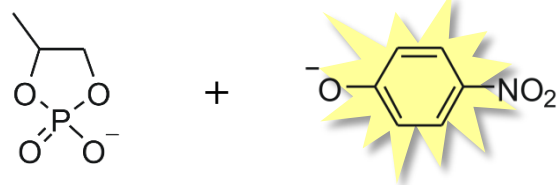
[TACN • Zn(II)] = 10 μ M
pH = 7.0, T = 25 $^{\circ}$ C.



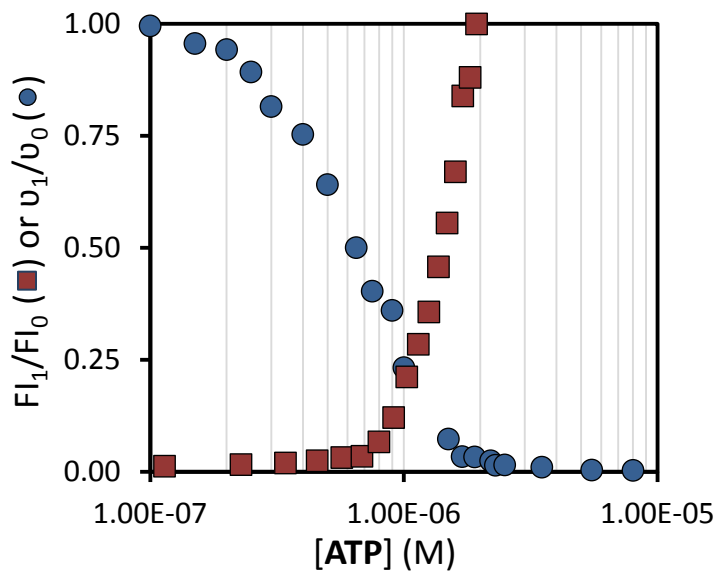
Catalytic signal generation



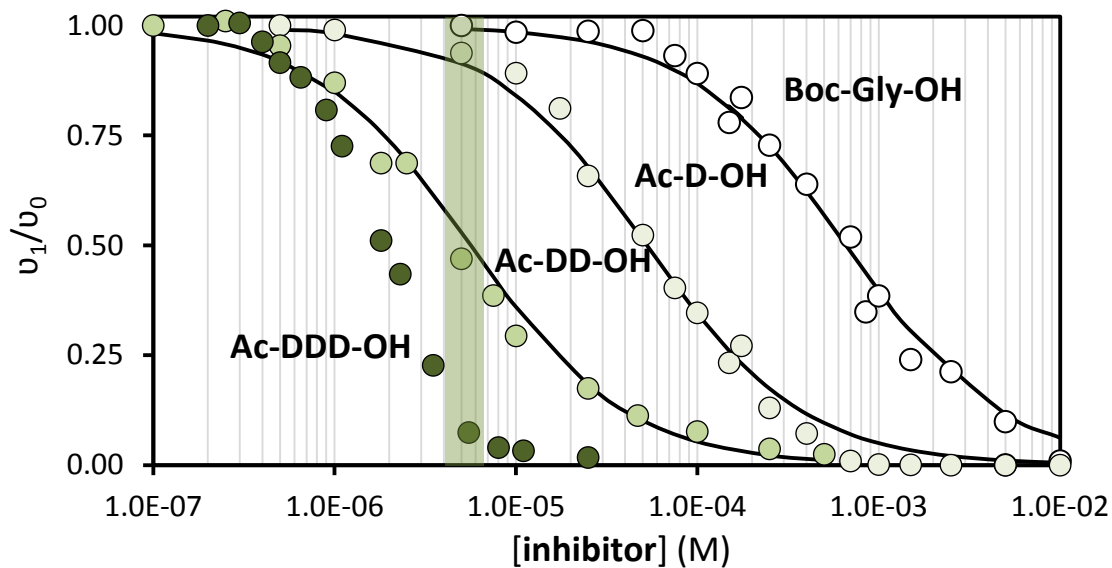
H₂O, pH 7, 40 °C



⊖ ⊖ ⊖ : ATP

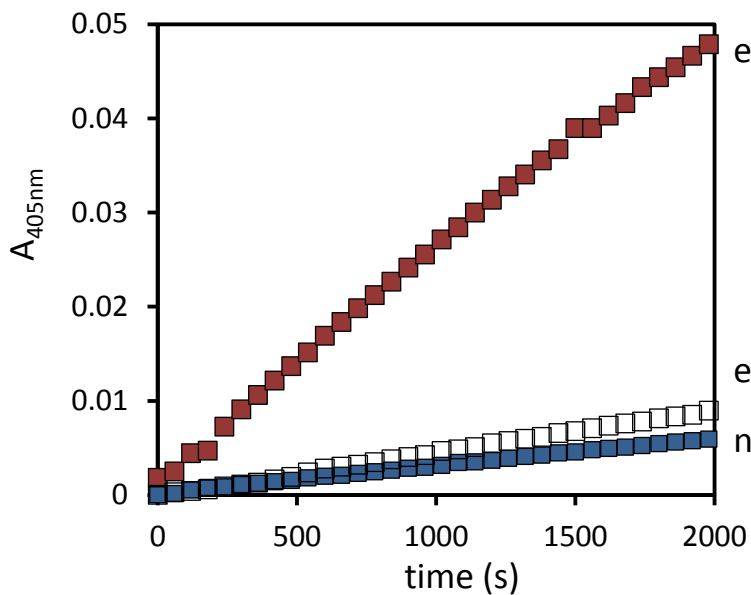
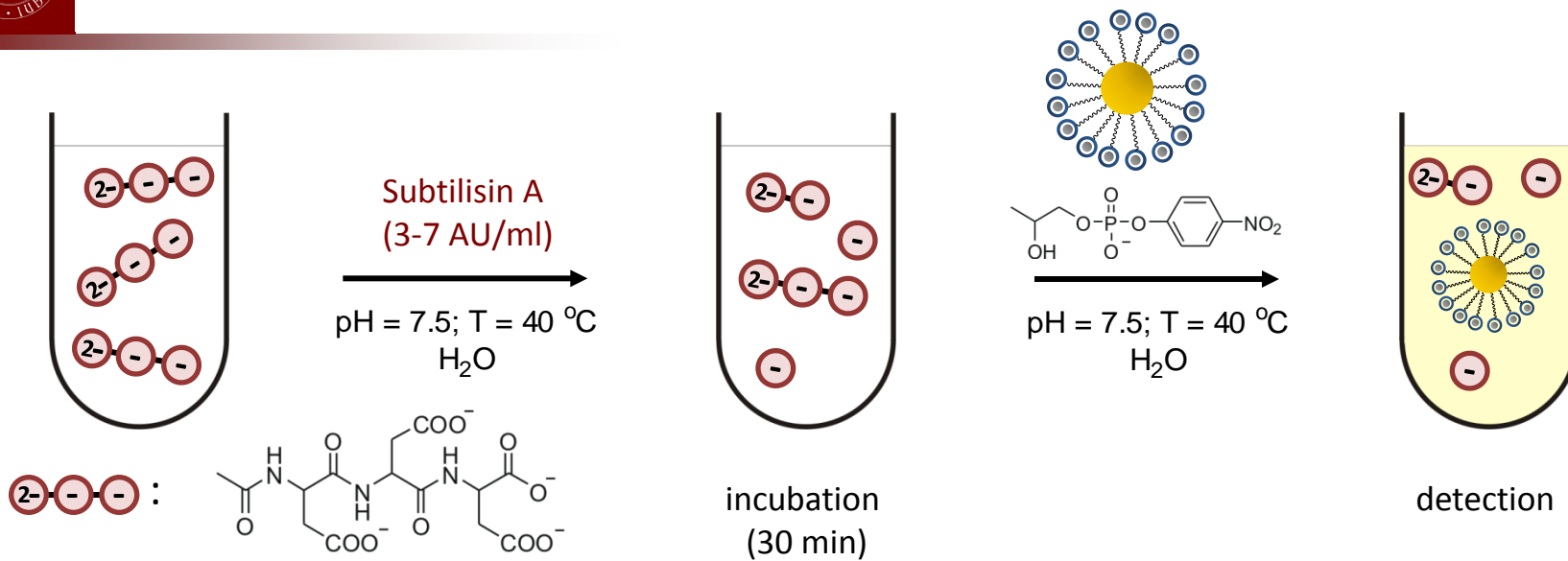


⊖ ⊖ ⊖ : peptides



[TACN • Zn(II)] = 5 μM, [substrate] = 1 mM, [HEPES] = 10 mM, pH = 7.0, T = 25 °C.

Enzyme assay

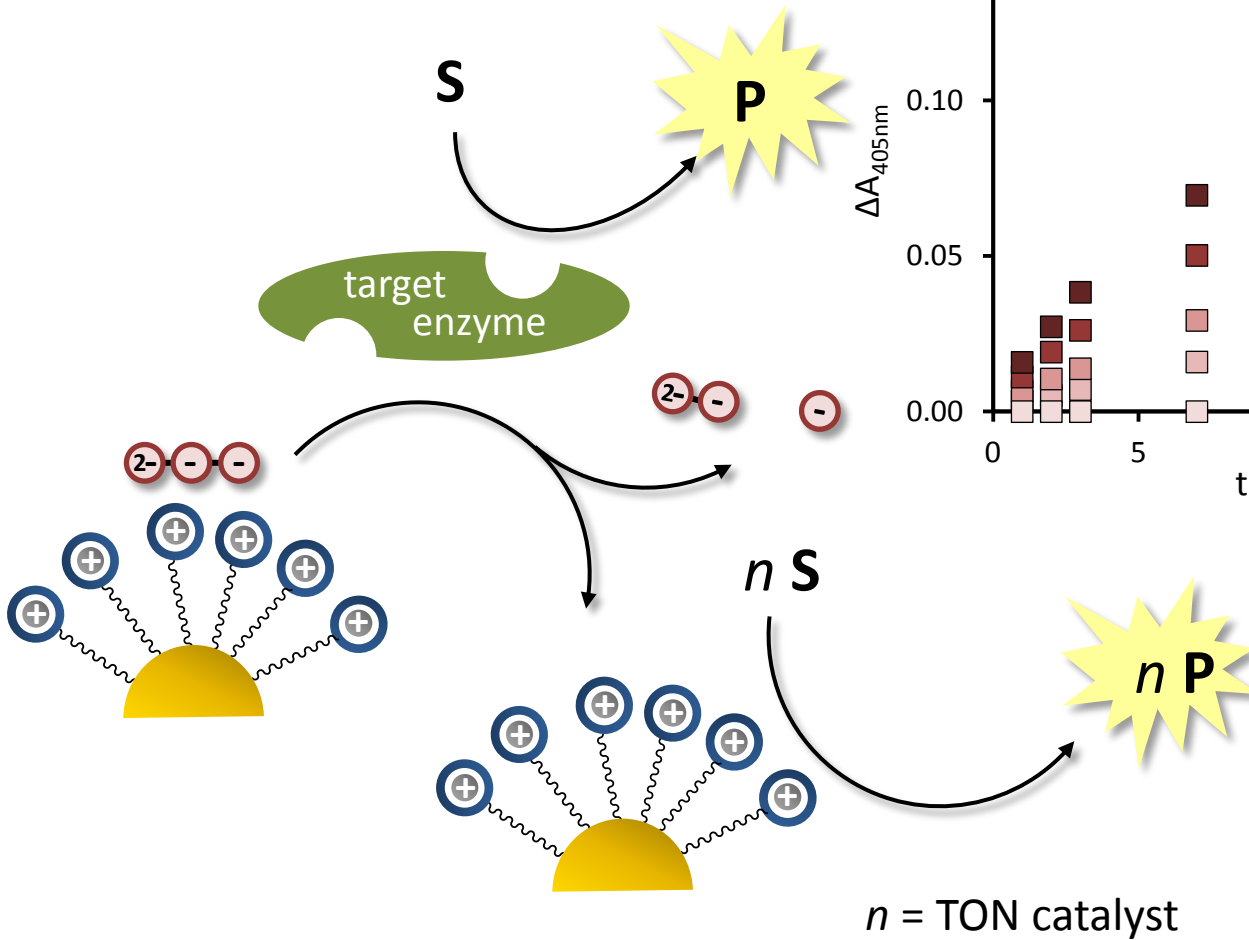


	enzyme
<chem>CC(=O)N[C@@H](CC(=O)N[C@@H](CC(=O)O)C(=O)O)C(=O)O</chem>	GCP II
<chem>CC(=O)N[C@@H](CC(=O)N[C@@H](C(C)C)C(=O)N[C@@H](CC(=O)O)C(=O)O)C(=O)O</chem>	Caspase-1

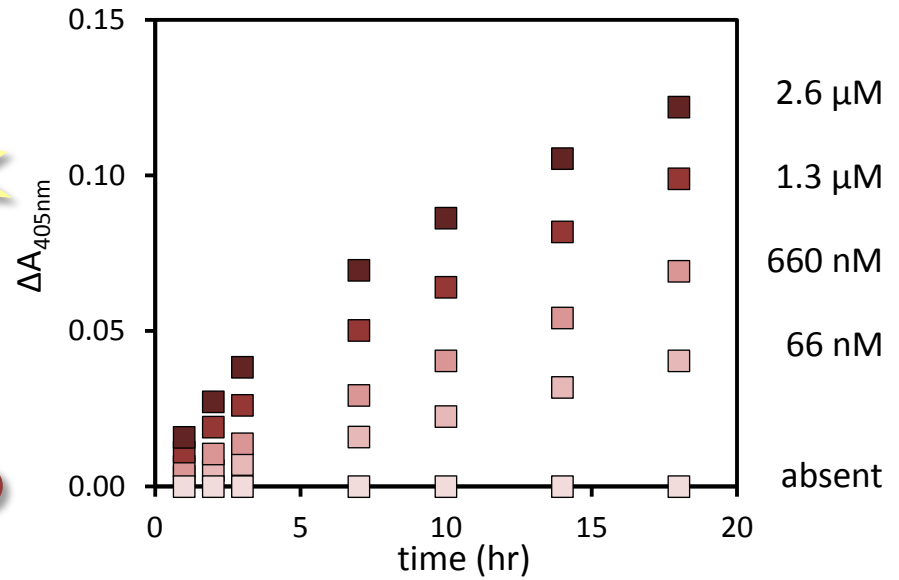


Sensing through signal amplification

mM
μ M
nM
pM
fM
aM
zM
yM



Subtilisin A

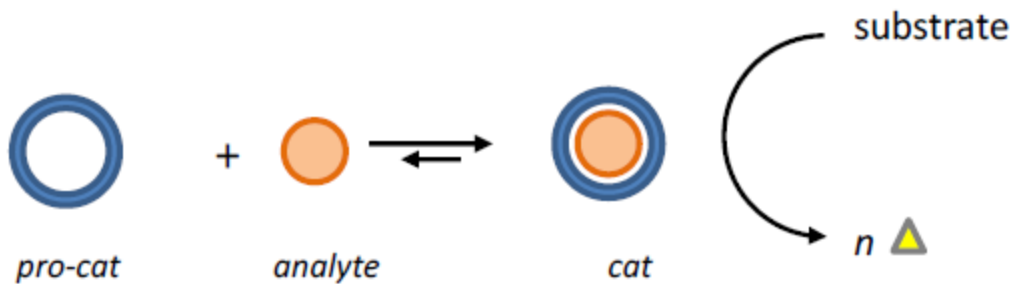


Importance: the analyte (enzyme) triggers a cascade of catalytic events



Supramolecular sensing systems

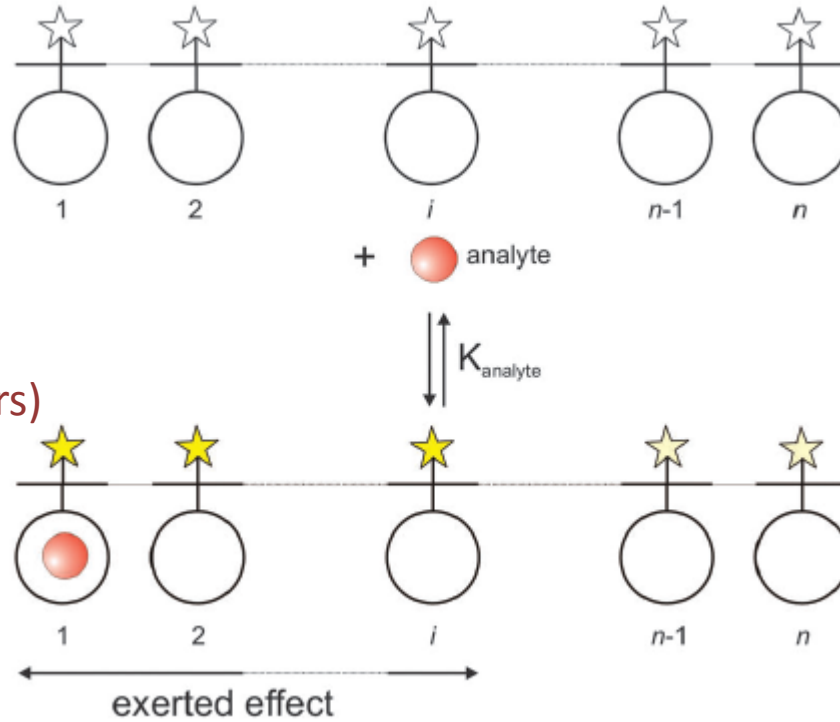
catalysis



multivalency



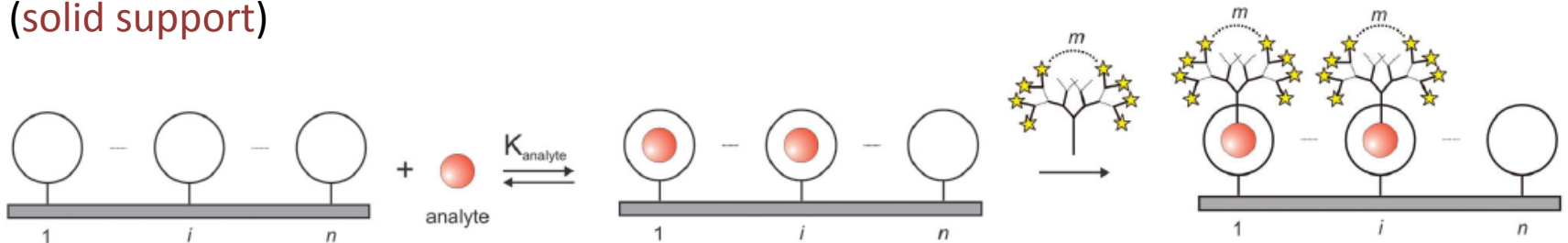
A



collective property (polymers)

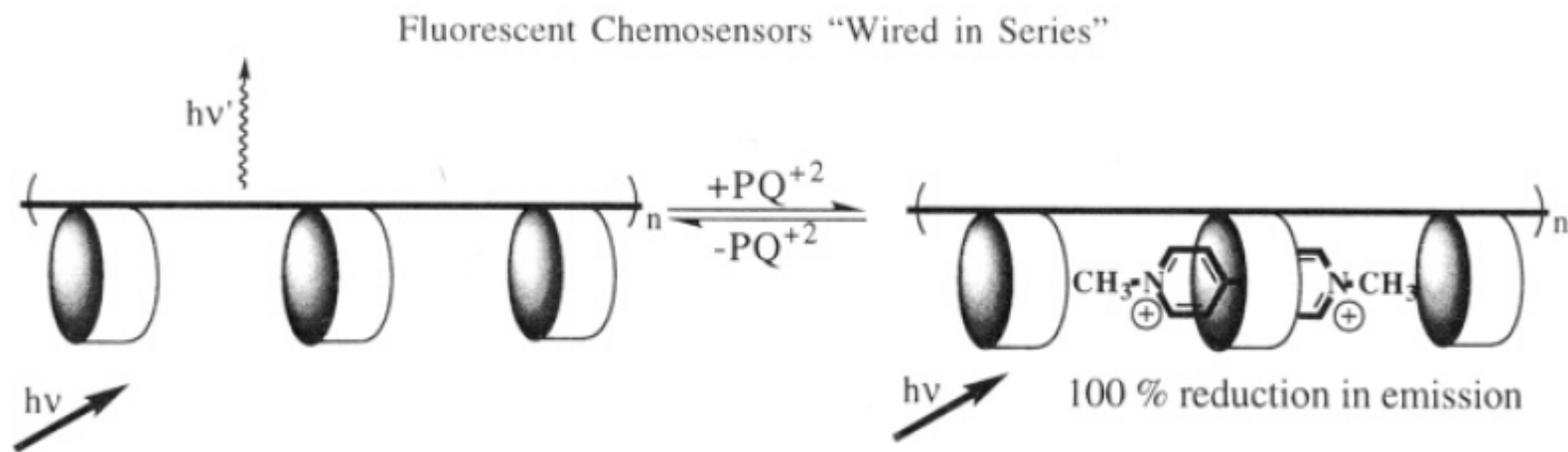
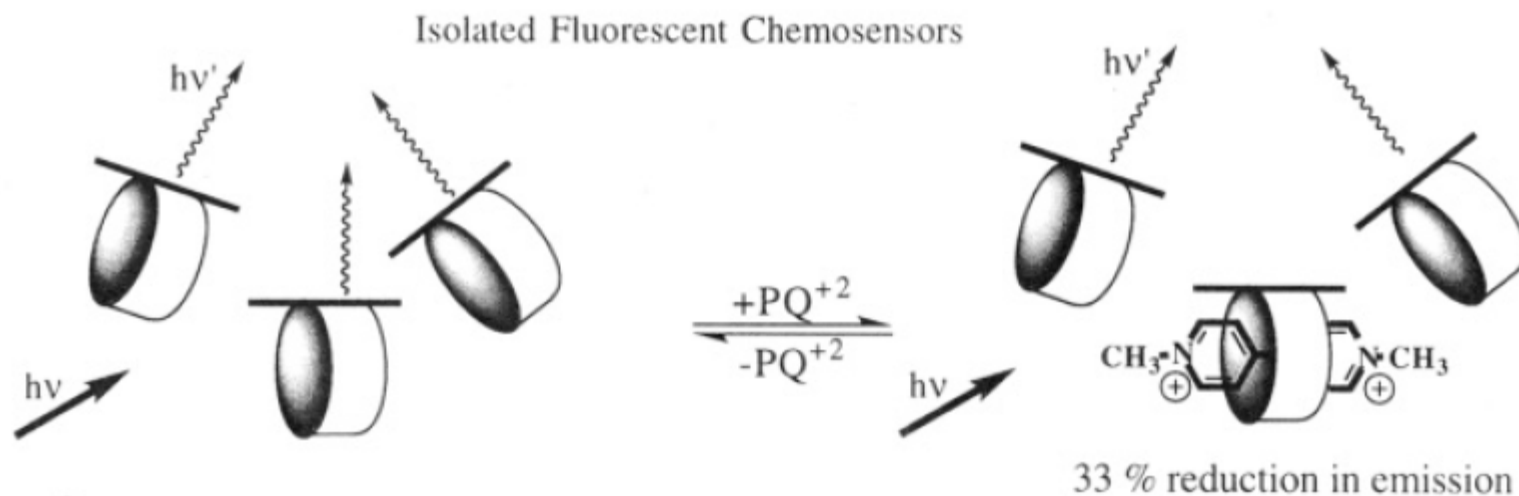
- electron conductivity
- helicity
- stability

B (solid support)

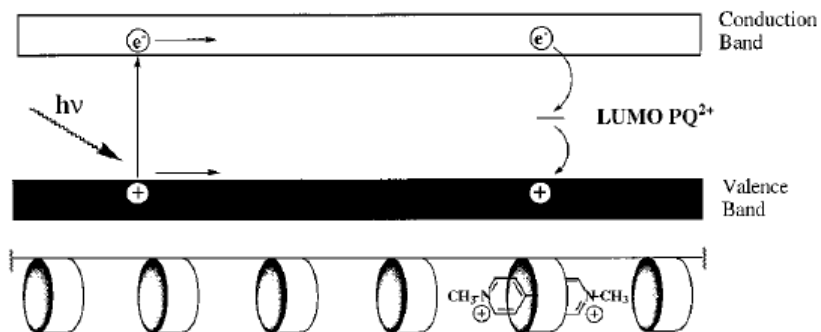




The 'Molecular Wire' approach



conjugated fluorescence polymer



Key issues:

- multitude of binding sites (guest binding at low concentrations)
- very high apparent binding constant
- polymerization degree
- mobility of the excited state
- fluorescence lifetime



Small molecule sensing

mM

μM

nM

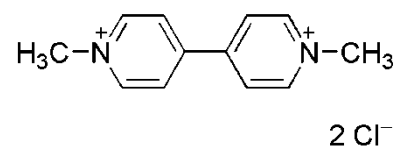
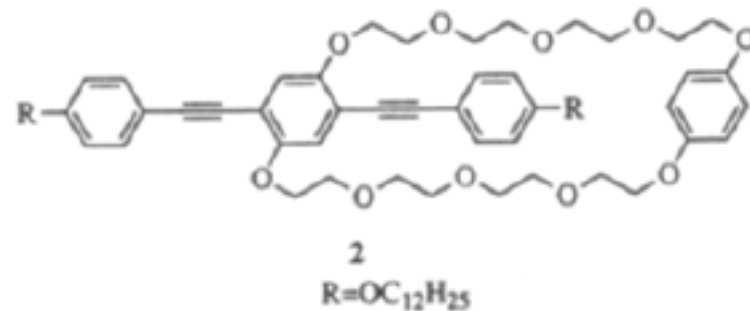
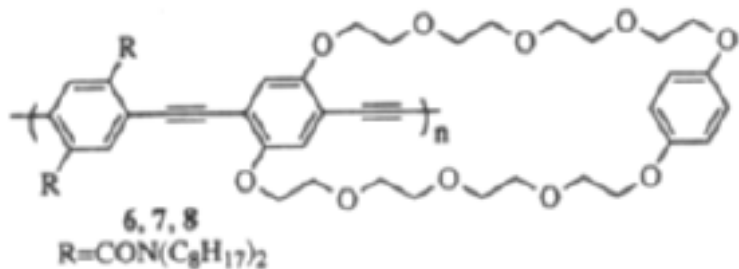
pM

fM

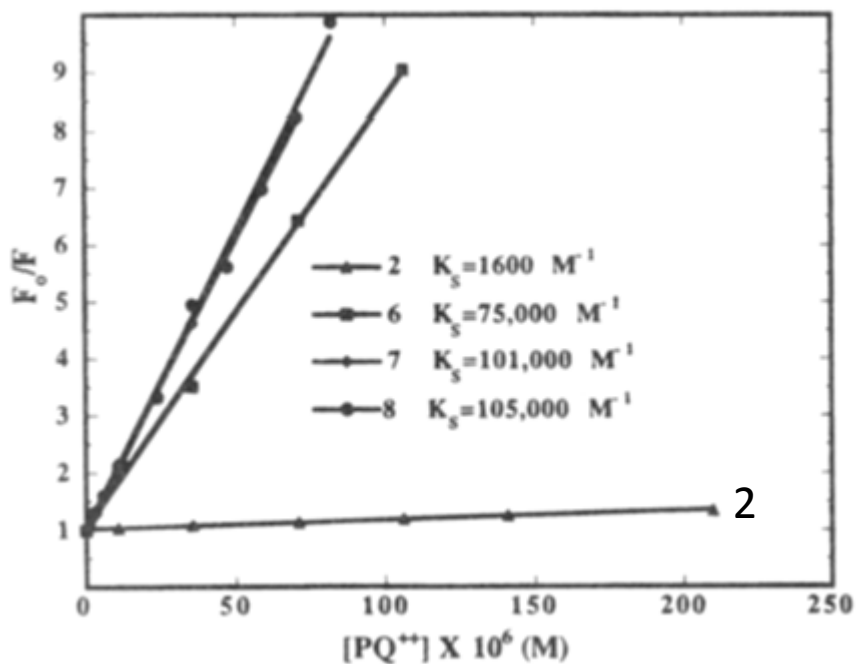
aM

zM

yM



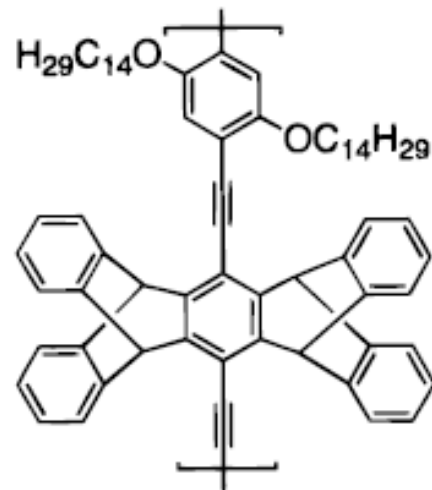
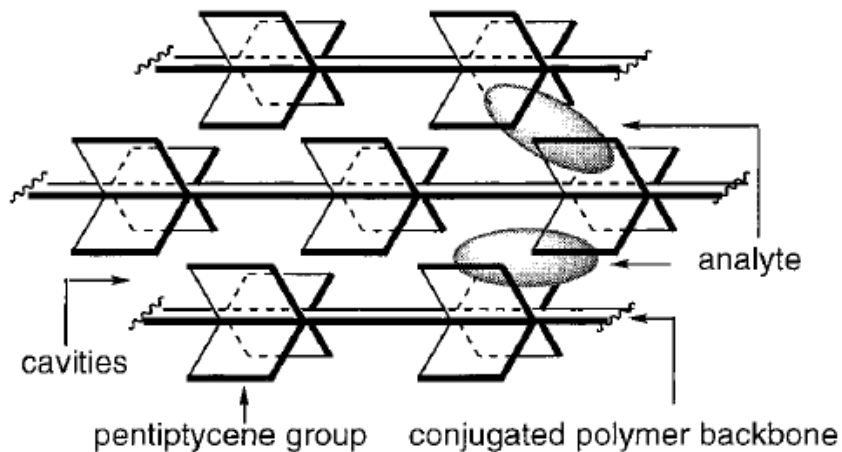
control



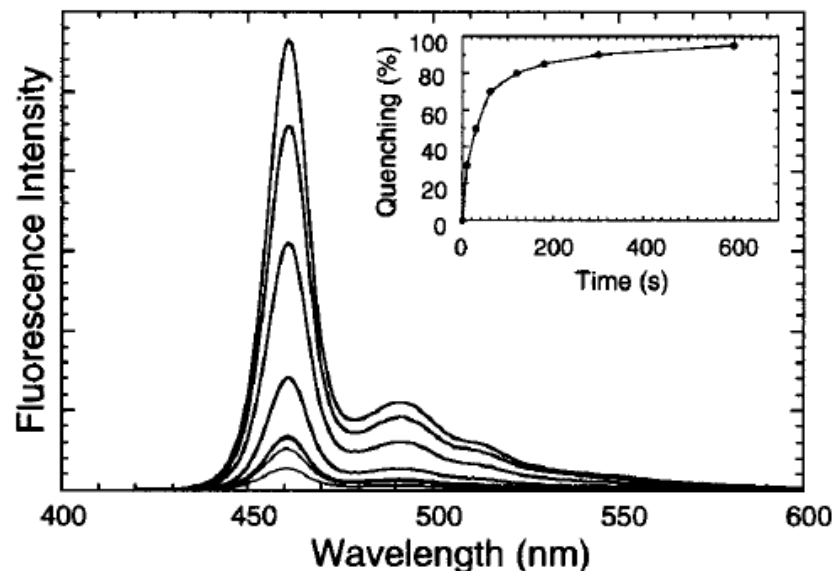
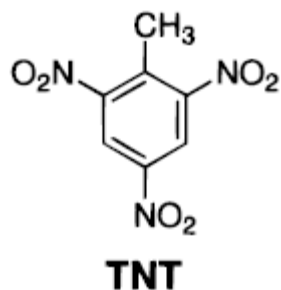
Mean molecular weight (PDI)

- 6: 31 100 (1.6)
- 7: 65 400 (1.6)
- 8: 122 500 (1.8)

Stern-Volmer plots ($F_0/F=1-K[PQ^{2+}]$)

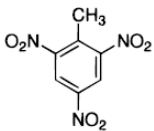
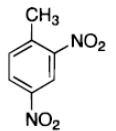
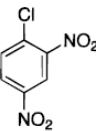
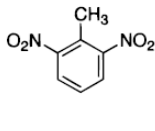
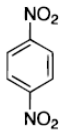
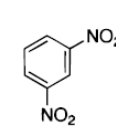
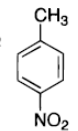
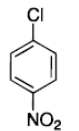
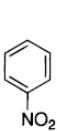


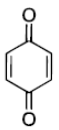
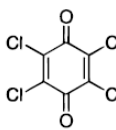
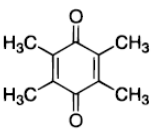
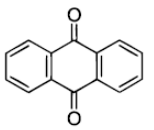
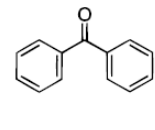
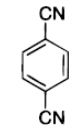
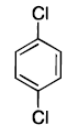
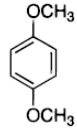
1. No π -stacking of polymer backbones, thus high fluorescence quantum yields and spectroscopic stability in thin films.
2. Second, reduced interpolymer interactions enhance the solubility
3. The cavities generated allow diffusion of small organic molecules into the films.

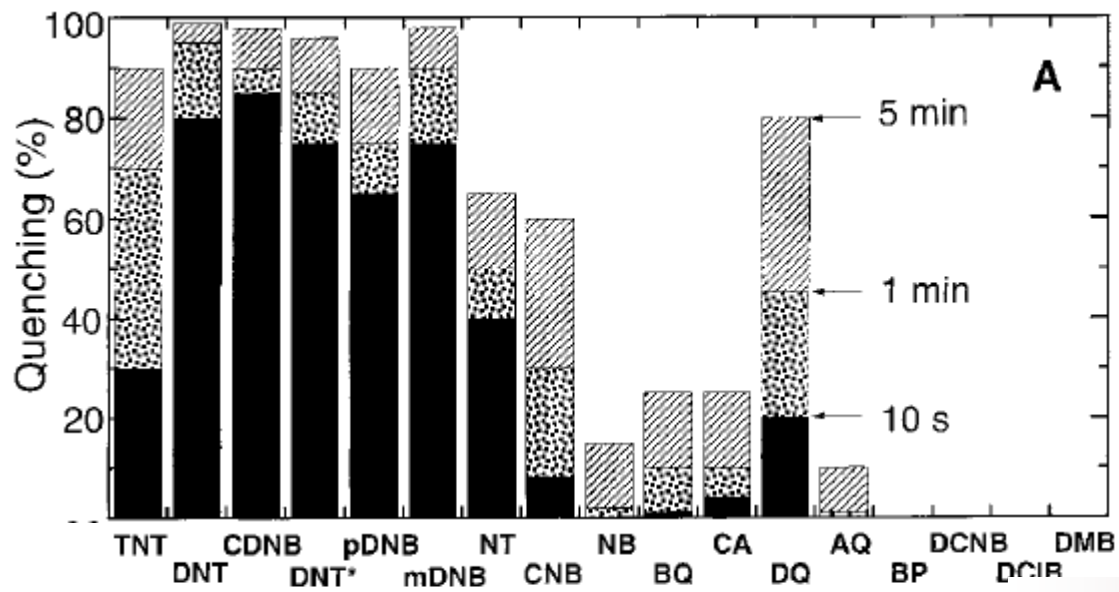




Fluorescent Porous Polymer Films as TNT Chemosensors

									
	TNT	DNT	CDNB	DNT*	pDNB	mDNB	NT	CNB	NB
rel. VP	1	18	10	70	3	110	2×10^4	3×10^3	3×10^4
E_{red} (V)	-0.7	-1.0	-0.8	-1.0	-0.7	-0.9	-1.2	-1.1	-1.15

								
	BQ	CA	DQ	AQ	BP	DCNB	DCIB	DMB
rel. VP	1×10^5	950	360	0.015	200	700	1×10^6	1×10^4
E_{red} (V)	-0.5	0.0	-0.8	-0.9	-1.6	-1.7	-1.8	< -2

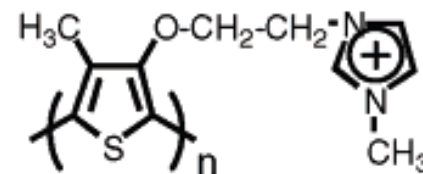


vapour pressure (VP) TNT
= 10 ppb at 25 °C

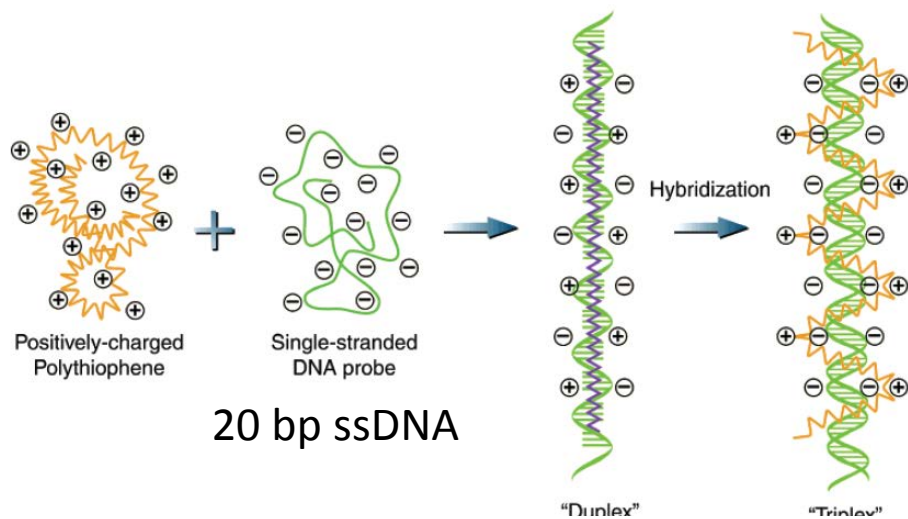




Fluorescent Polymers for DNA detection

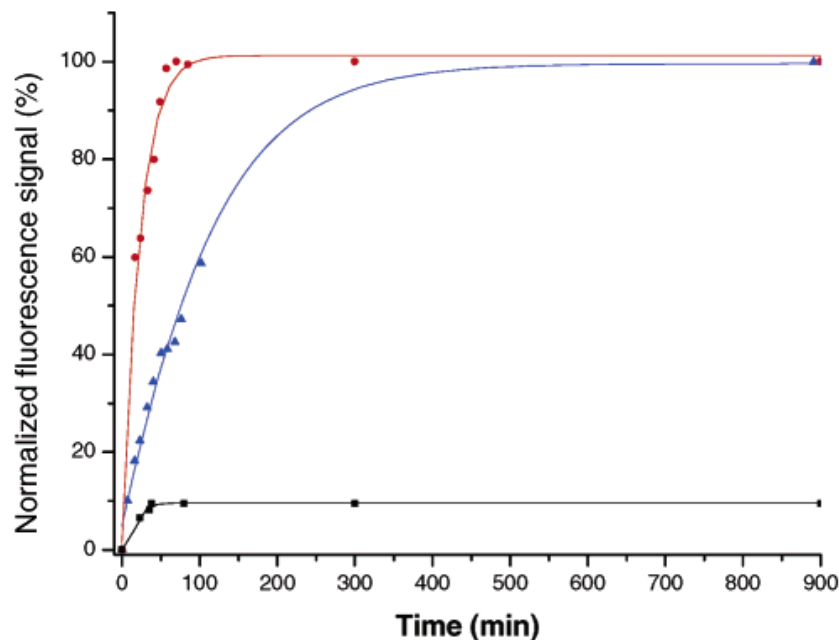


6 - 10 kDa (PDI:1.2-2.9)



duplex form/aggregate
decreased fluorescence

triplex form
increased fluorescence



red: target

blue: one mismatch

black: double mismatch

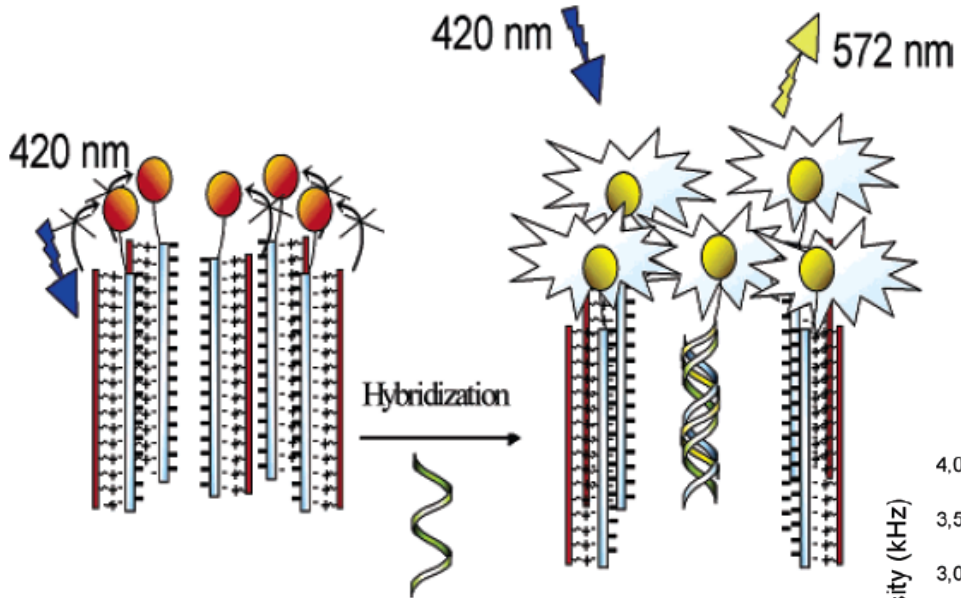
The limit of detection for the perfectly matched 20-mer target is 310 molecules, or 0.54 zmol, in a volume of 150 μ L.

mM
μ M
nM
pM
fM
aM
zM
yM



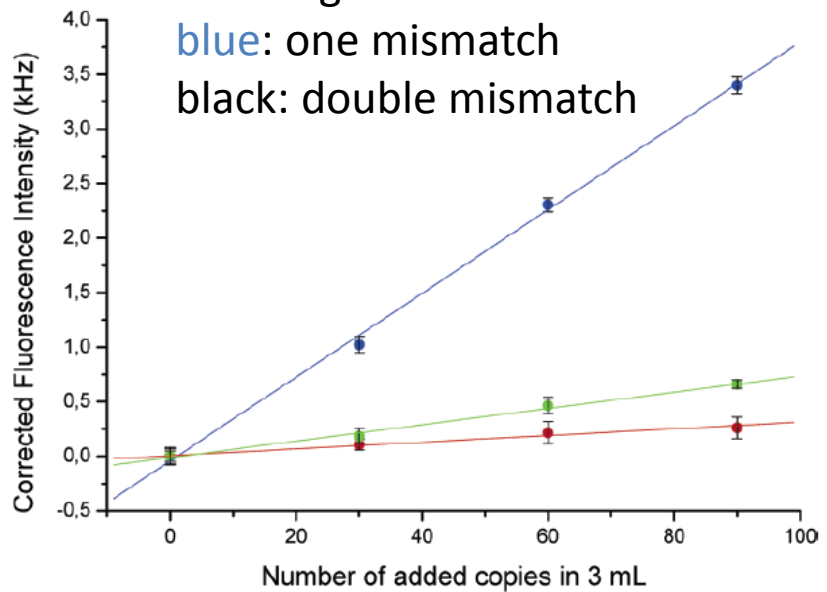
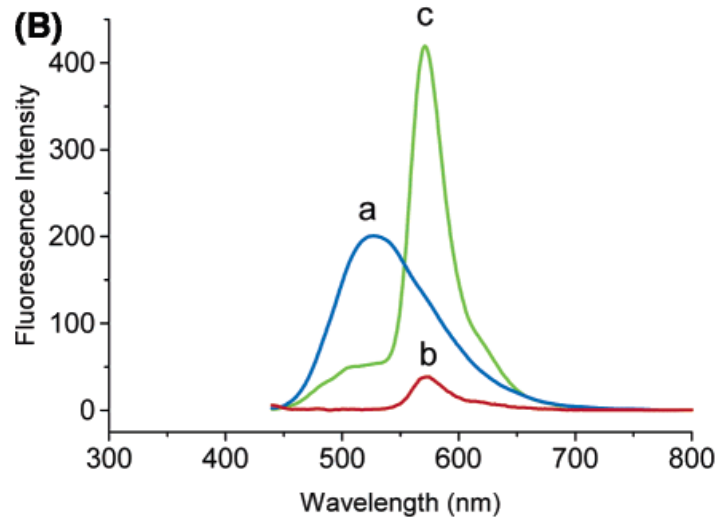
Ultrasensitive detection using fluorescence chain reaction (FCR)

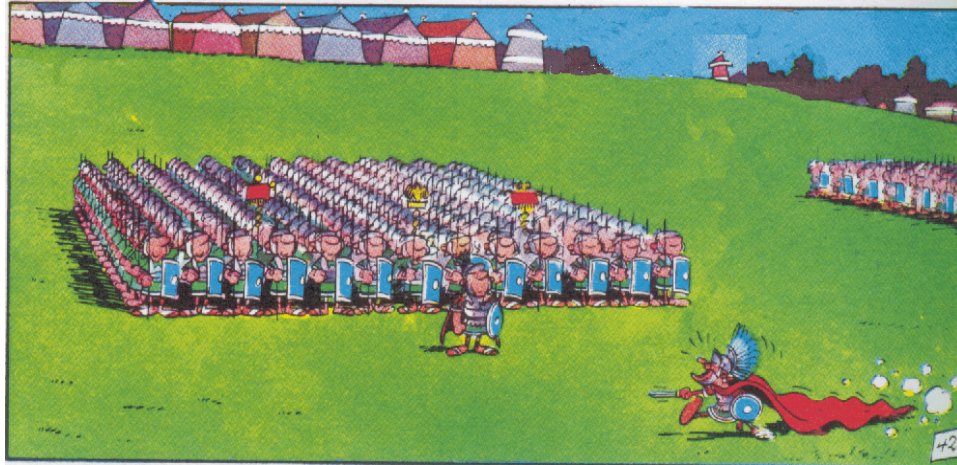
mM
μM
nM
pM
fM
aM
zM
yM



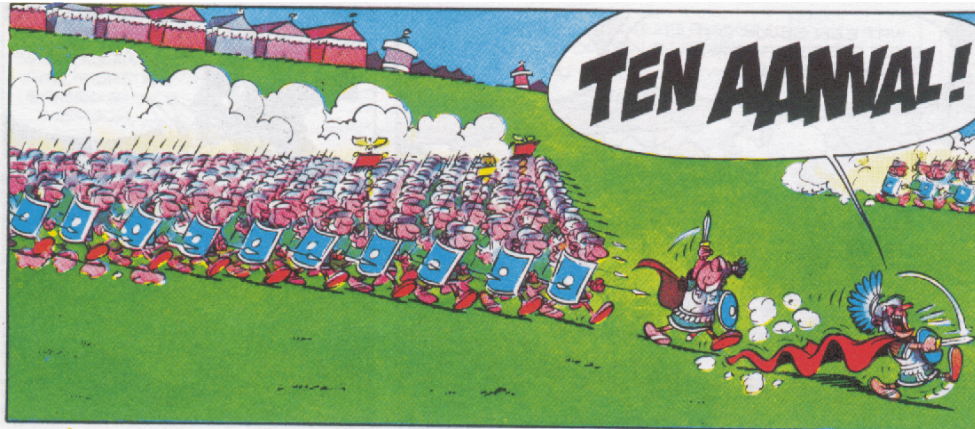
capture probe is labeled with a fluorophore (Alexa Fluor 546)
large number of capture probes are available ($\times 10^{10}$ copies)

red: target
blue: one mismatch
black: double mismatch





carré



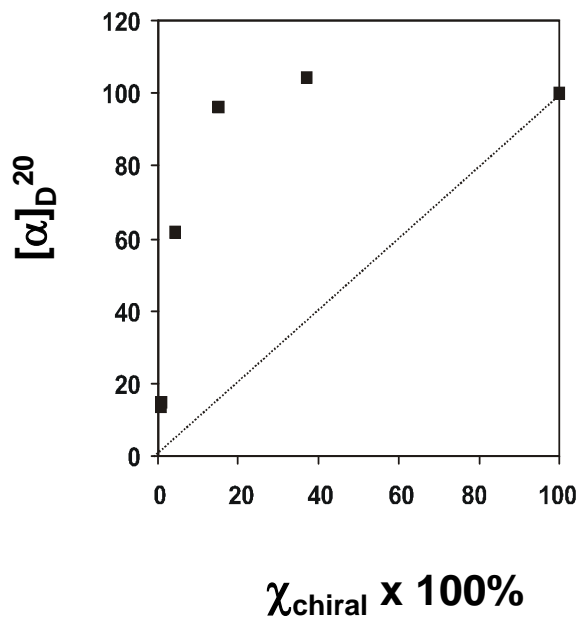
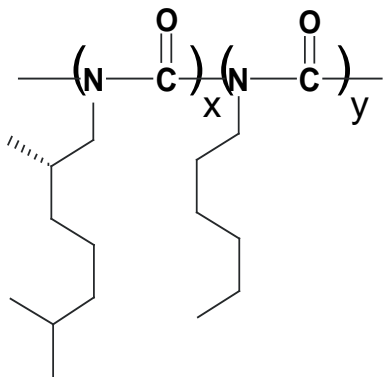
triangle



circle !



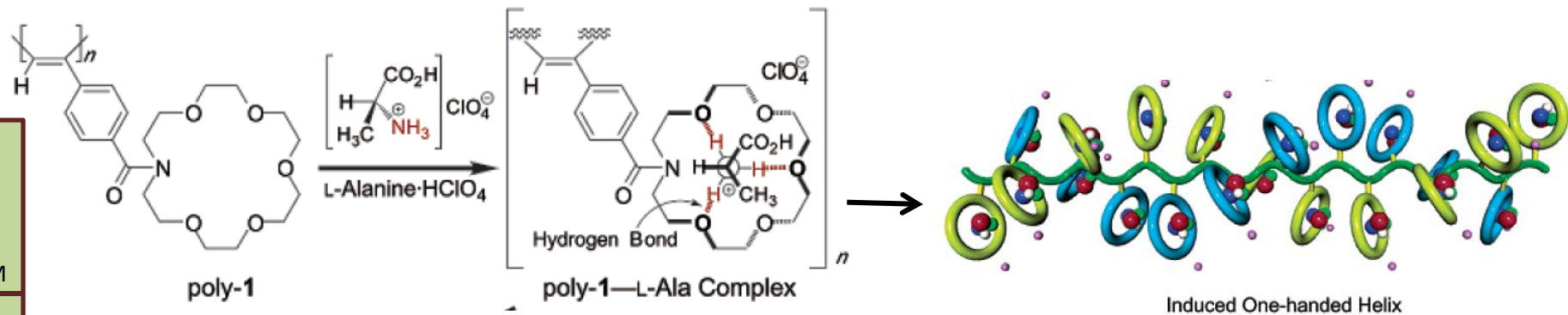
The 'Sergeants-and-Soldiers' principle in polymers





The use of 'Sergeants-and-Soldiers' for sensing

- mM
- μM
- nM
- pM
- fM
- aM
- zM
- yM





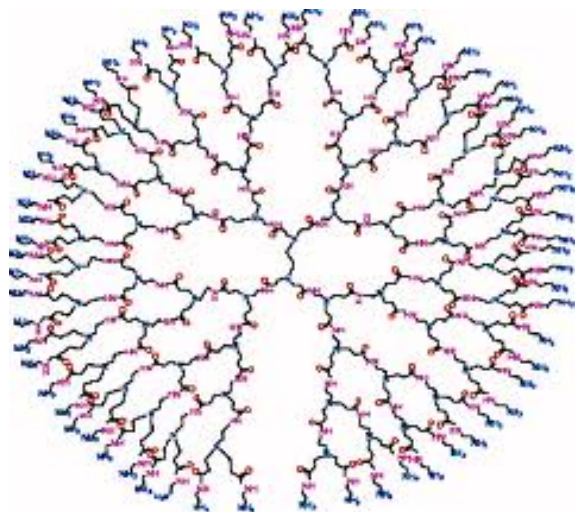
Detection of small enantiomeric excesses



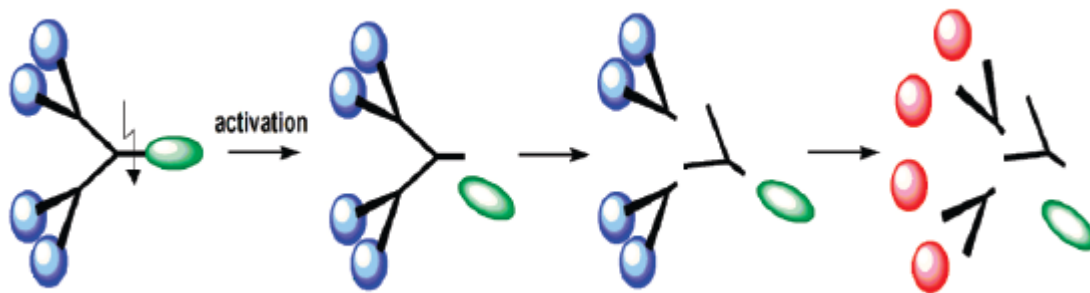
majority rules



Affecting macromolecule stability



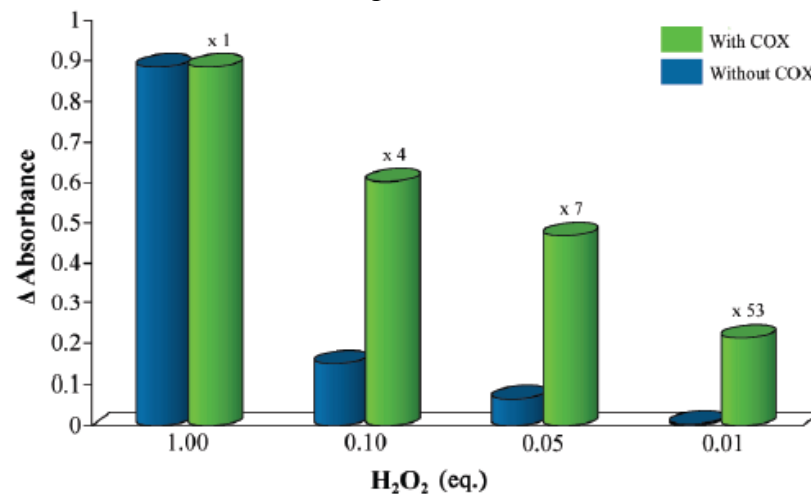
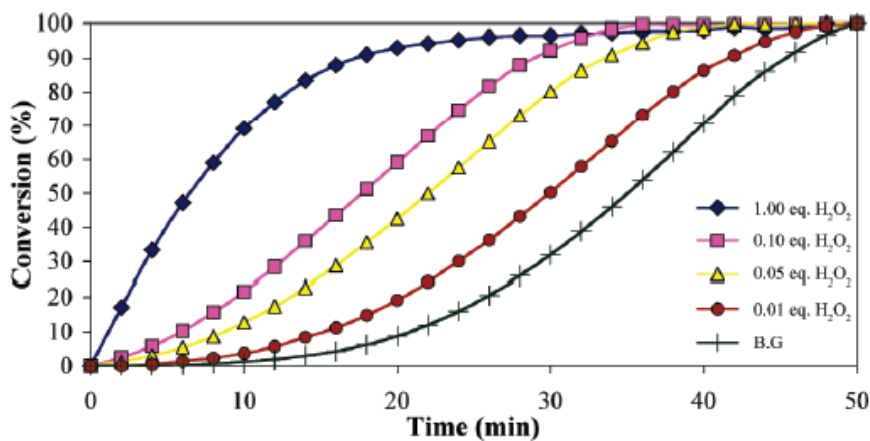
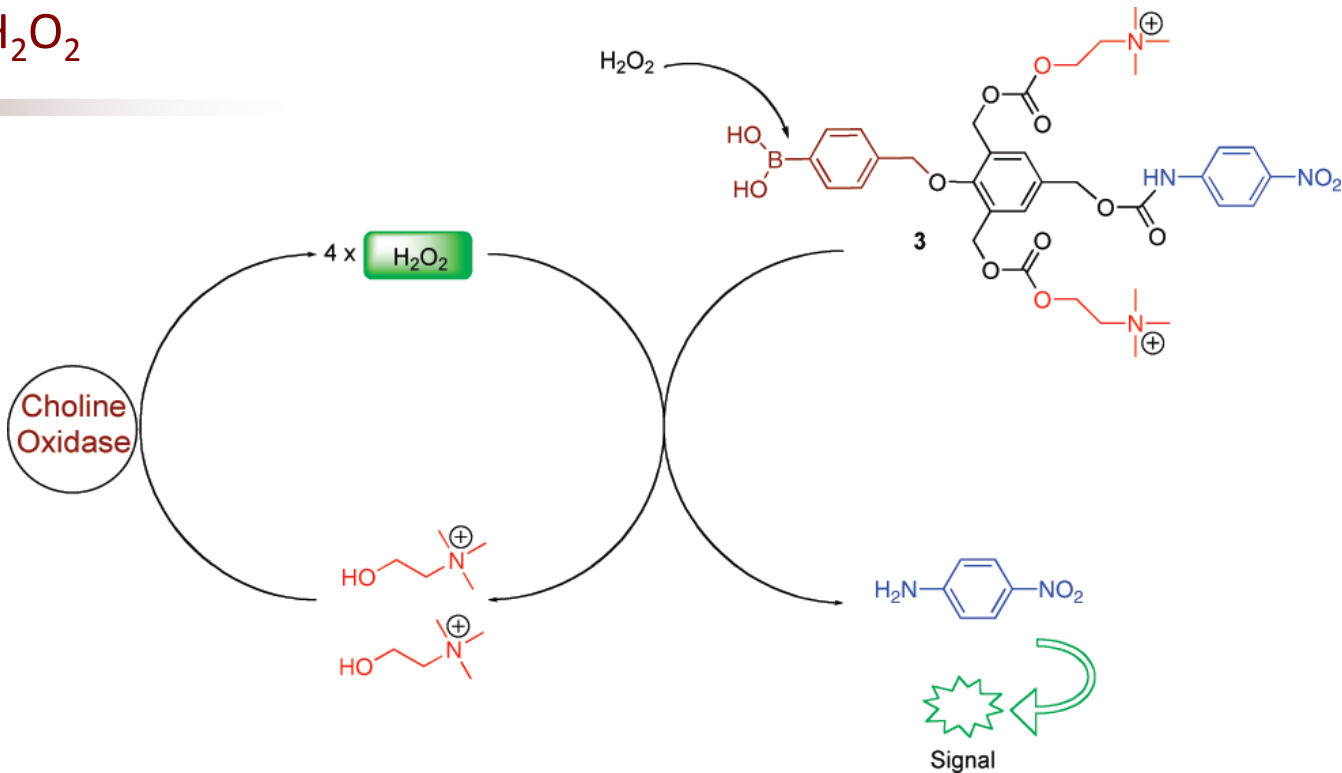
Dendrimers are repetitively branched molecules. The name comes from the Greek word "δένδρον" (pronounced dendron), which translates to "tree".



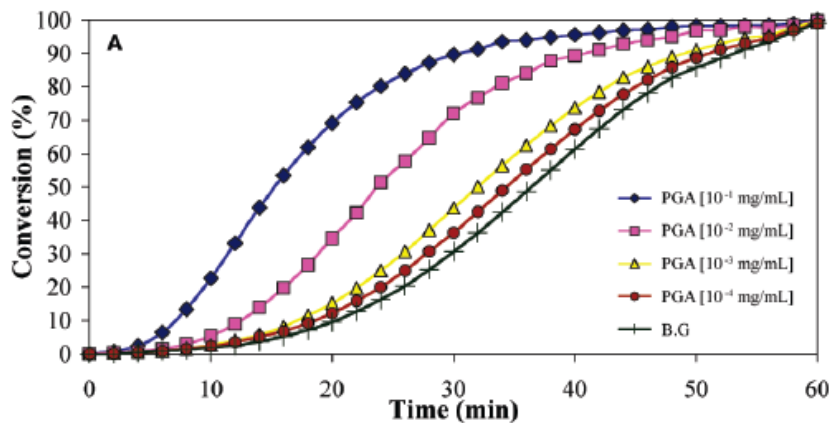
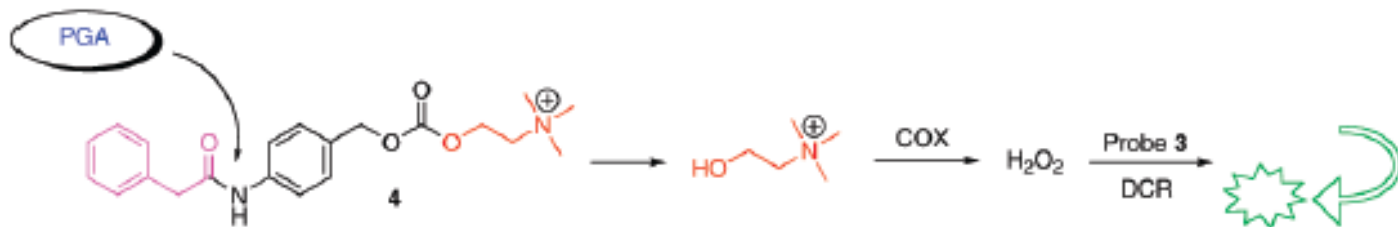


Sensing of H₂O₂

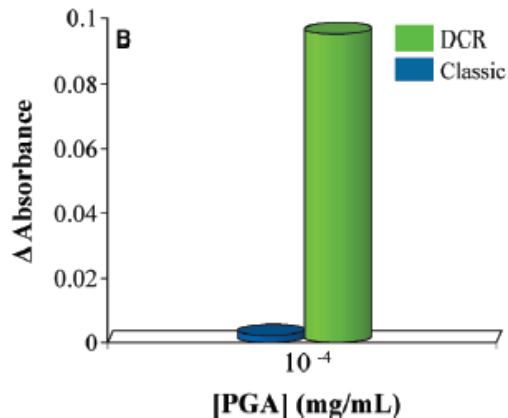
mM
μM
nM
pM
fM
aM
zM
yM



Adaptation for enzyme detection

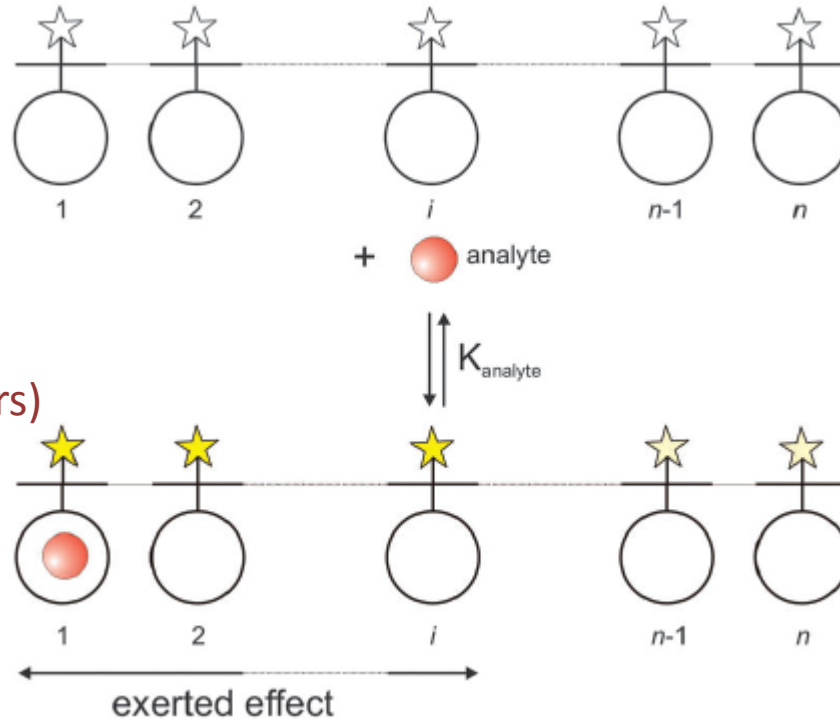


PGA
penicillin-G-amidase



Critical issue
intrinsic stability of the probe

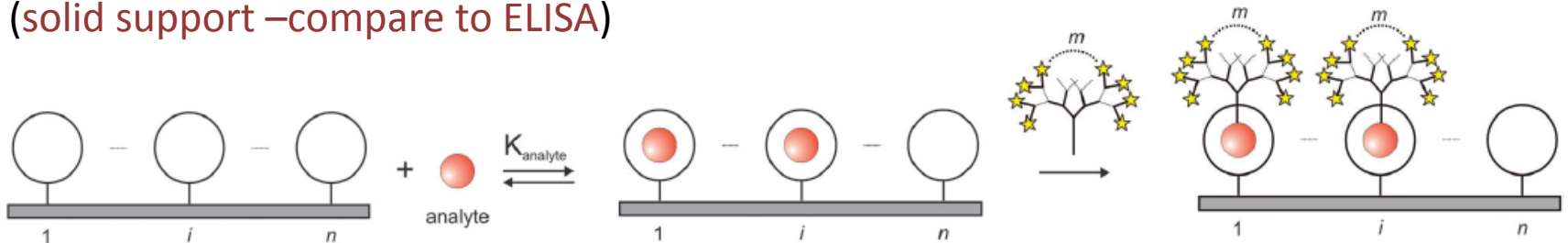
A



collective property (polymers)

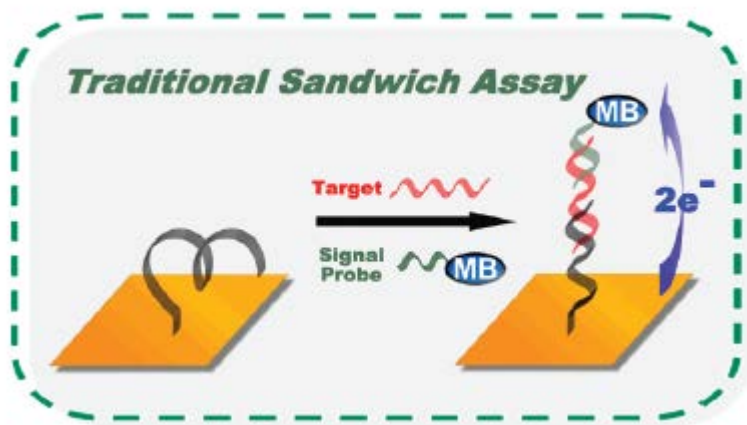
- electron conductivity
- helicity
- stability

B (solid support –compare to ELISA)

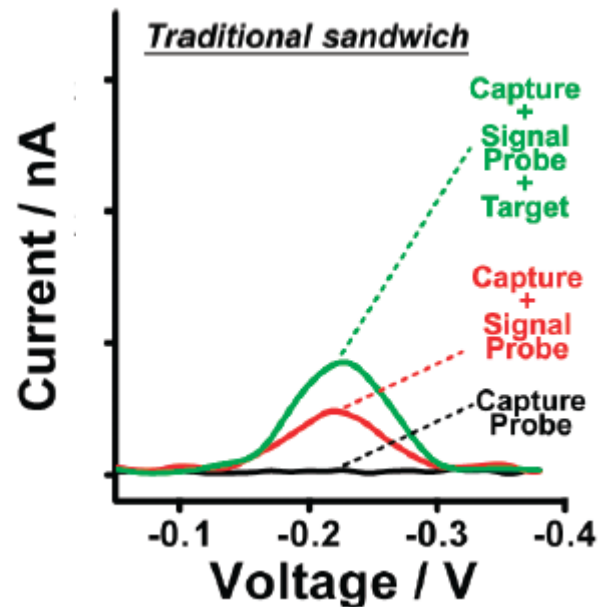
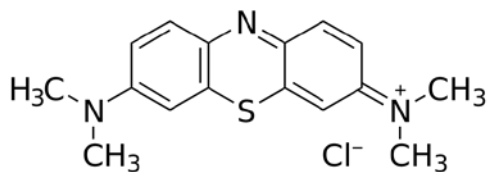


Sandwich assay for oligonucleotide detection

mM
μ M
nM
pM
fM
aM
zM
yM



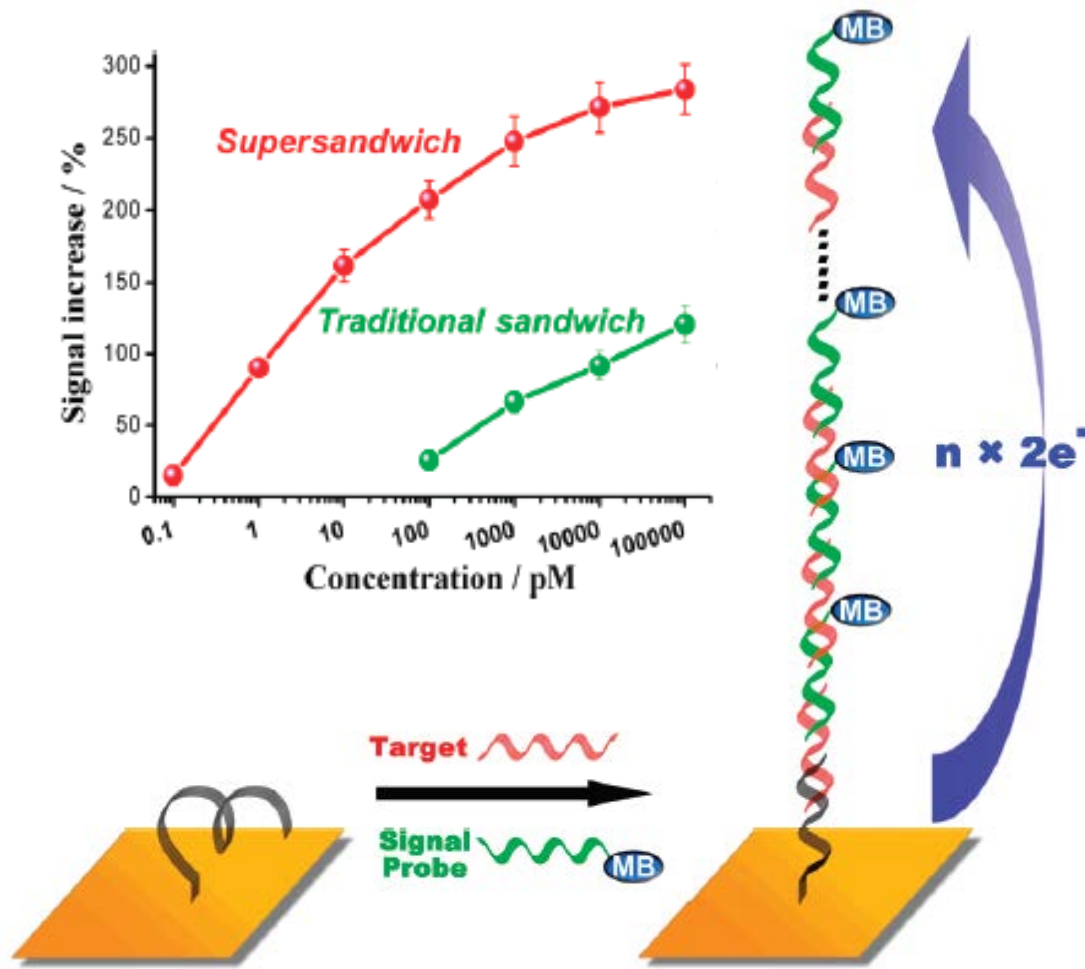
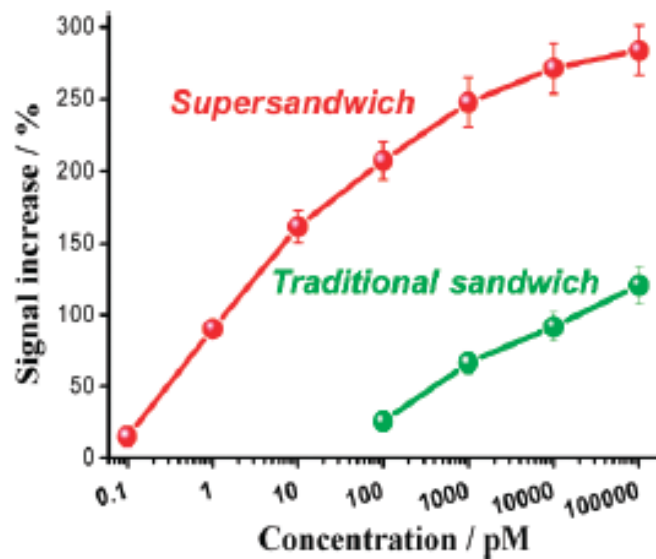
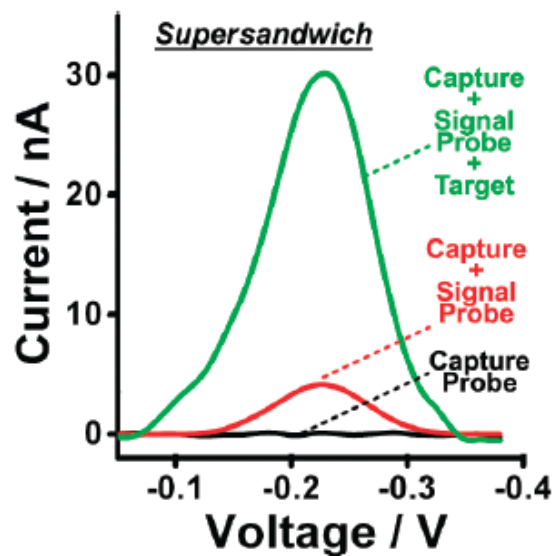
MB: methylene blue



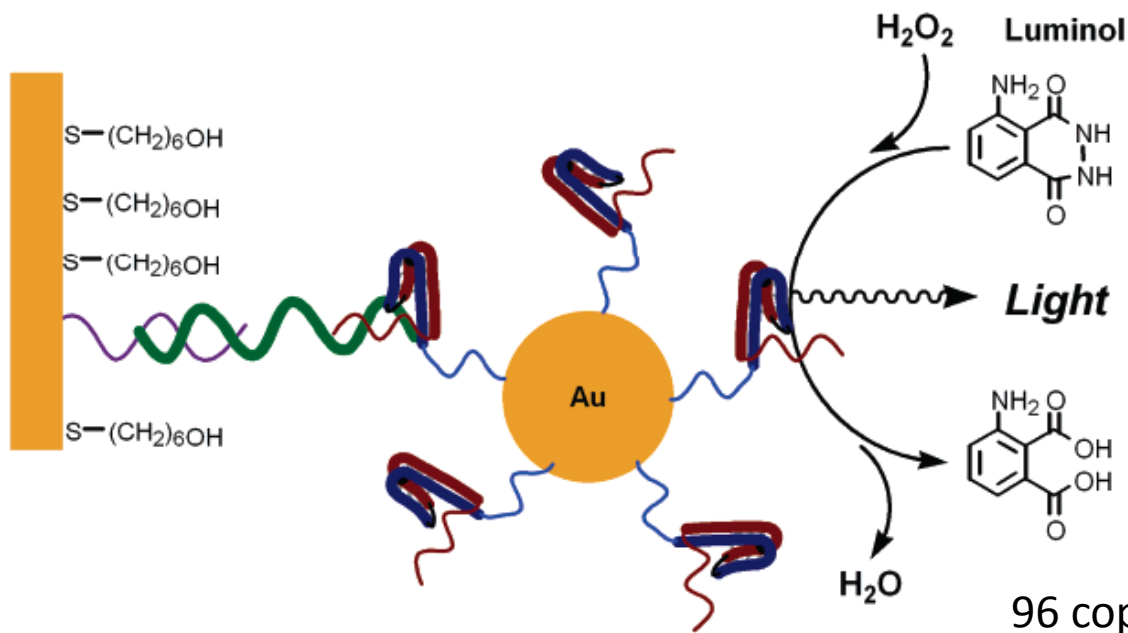
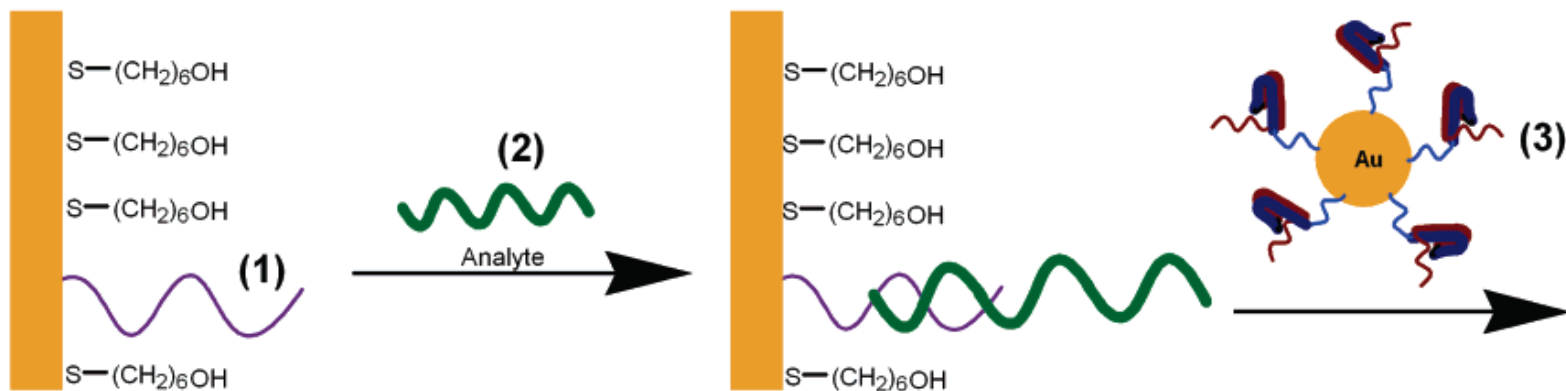


A Supersandwich assay for oligonucleotide detection

mM
μ M
nM
pM
fM
aM
zM
yM



DNAzyme-functionalized Au NPs for the amplified detection of DNA



signal amplification

- 1) multivalency
- 2) catalysis

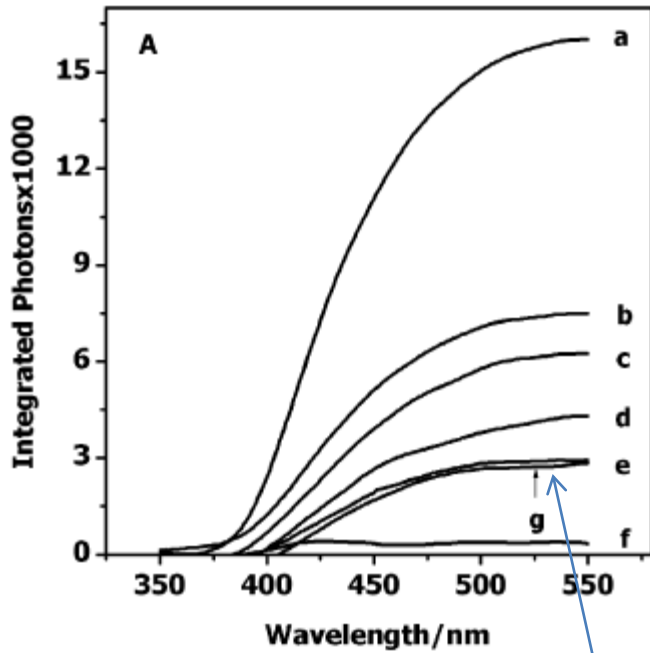
96 copies

(1) = 5'-HS(CH₂)₆CGATTCGGTACTGG-3'
 (2) = 5'-TTGAGCATGCGCATTATCTGAGCCAGTACCGAATCG-3'
 (3) = 5'-ATGCGCATGCTCAAT₁₀GGGTAGGGCGGGTTGGGT₁₇(CH₂)₆SH-3'
 (3a) = 5'-ATGCGCATGCTCAATTTGGGTAGGGCGGGTTGGG-3'



Response curves

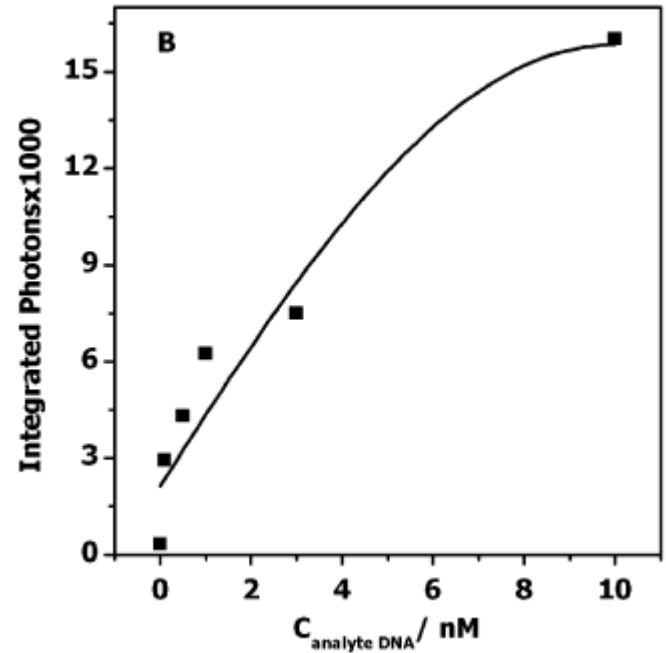
- mM
- μ M
- nM
- pM
- fM
- aM
- zM
- yM



reference without NP

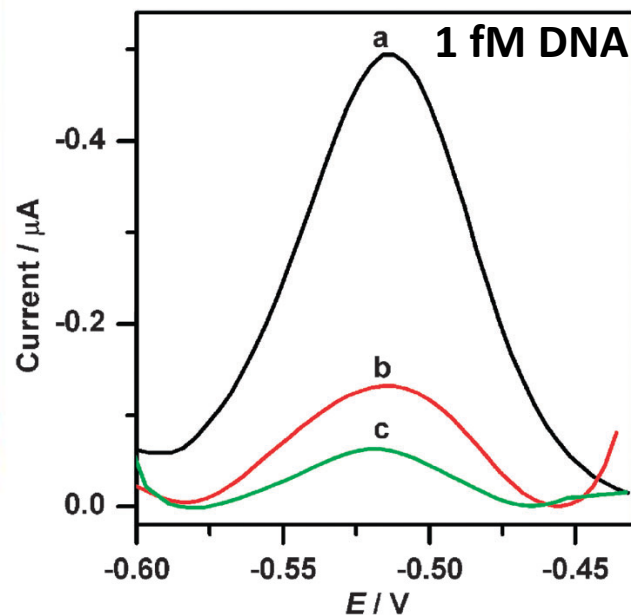
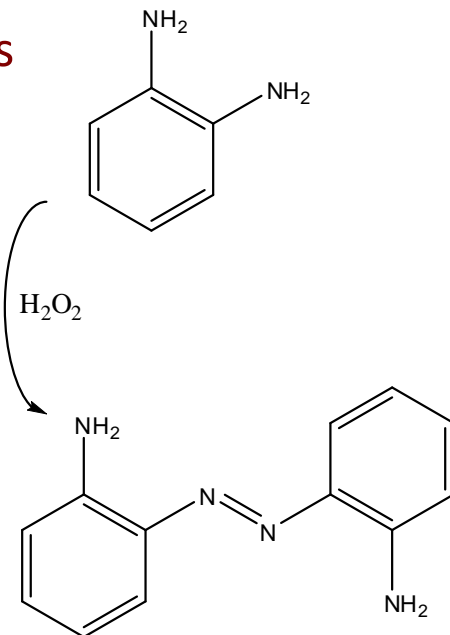
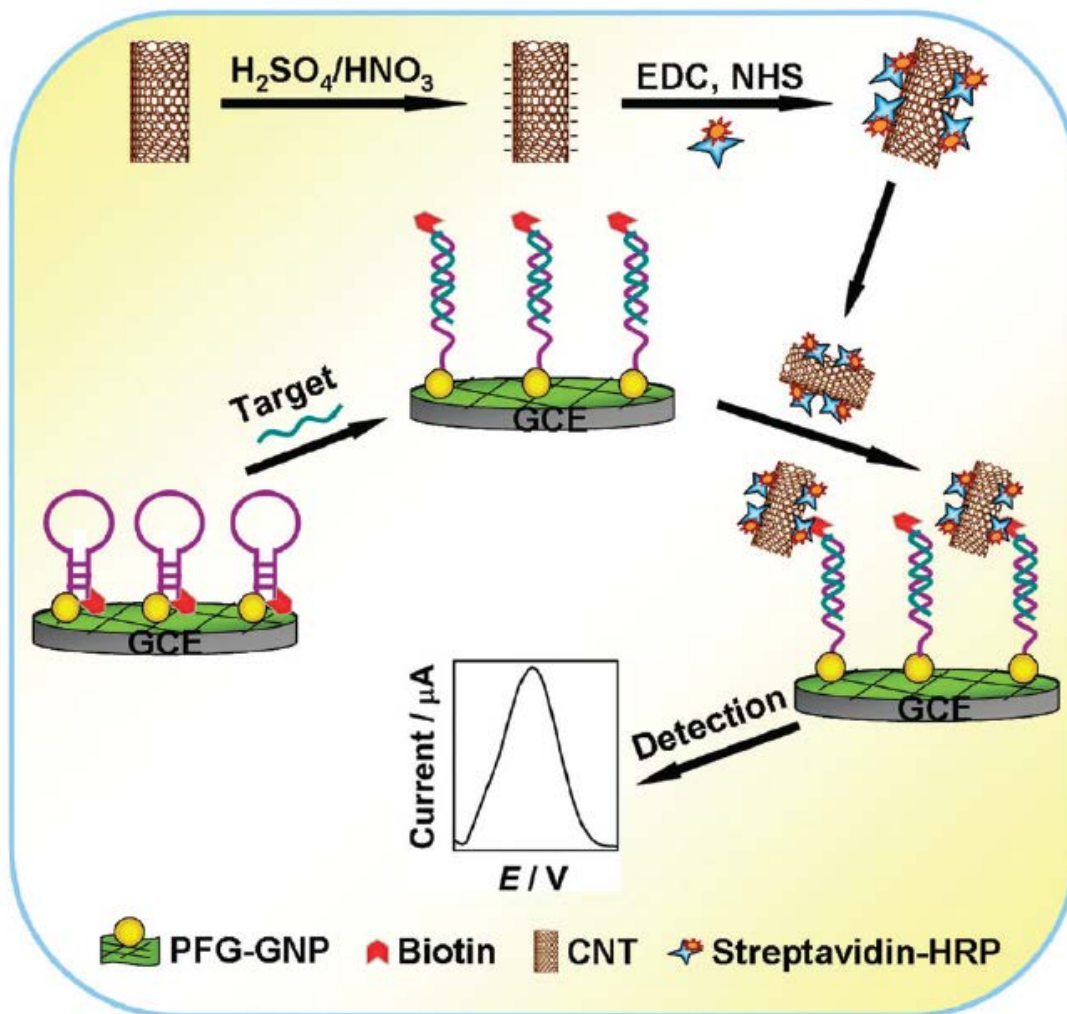
NPs cause 10-fold increase in sensitivity

calibration curve





Signal amplification using multivalent CNT-HRP constructs

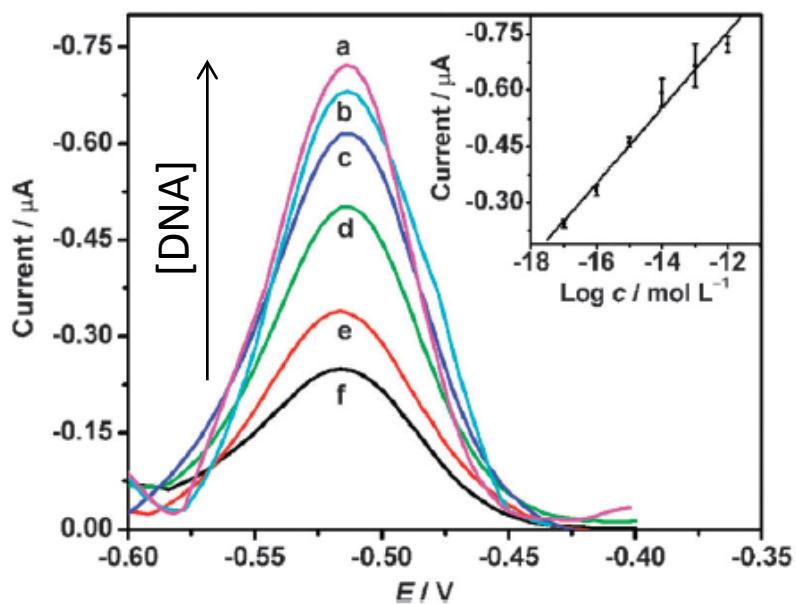


a) streptavidin-HRP-CNTs probe; b) streptavidin-HRP; c) streptavidin-HRP-CNTs probe (no DNA)

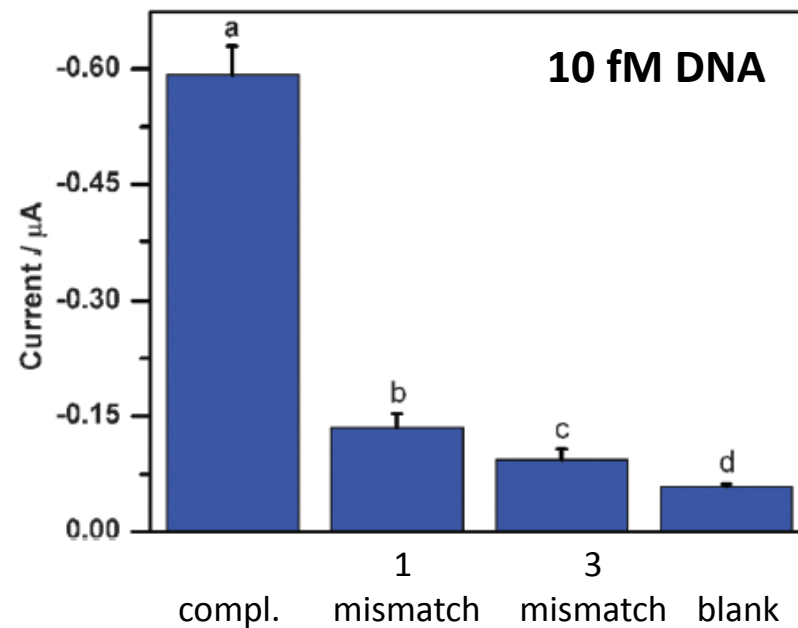


Amperometric detection of DNA

mM
μ M
nM
pM
fM
aM
zM
yM

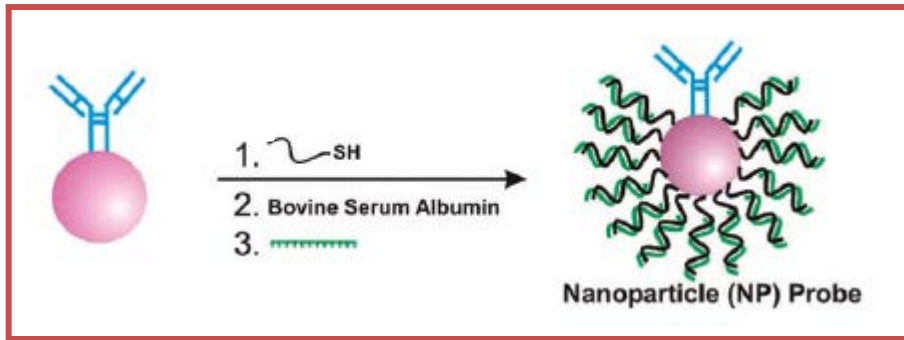


a = 1 pM
b = 100 fM
c = 10 fM
d = 1 fM
e = 100 aM
f = 10 aM

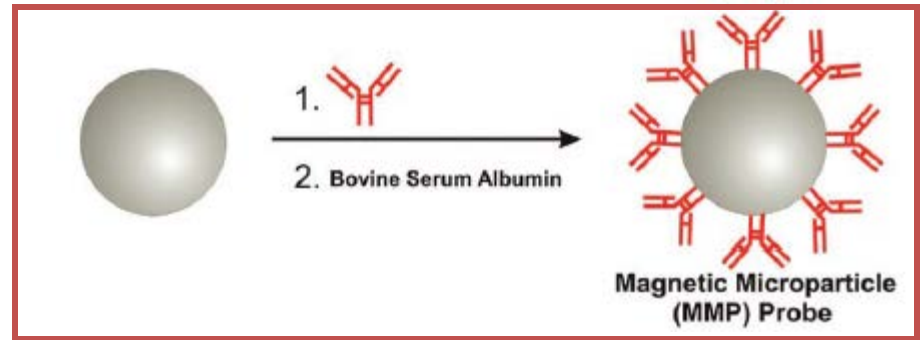




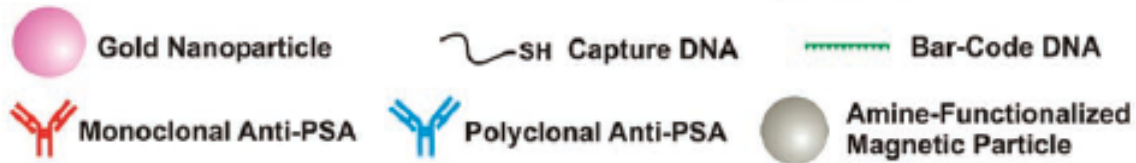
Bio-bar Codes (assay components)



NP: 13 - 30 nm diameter

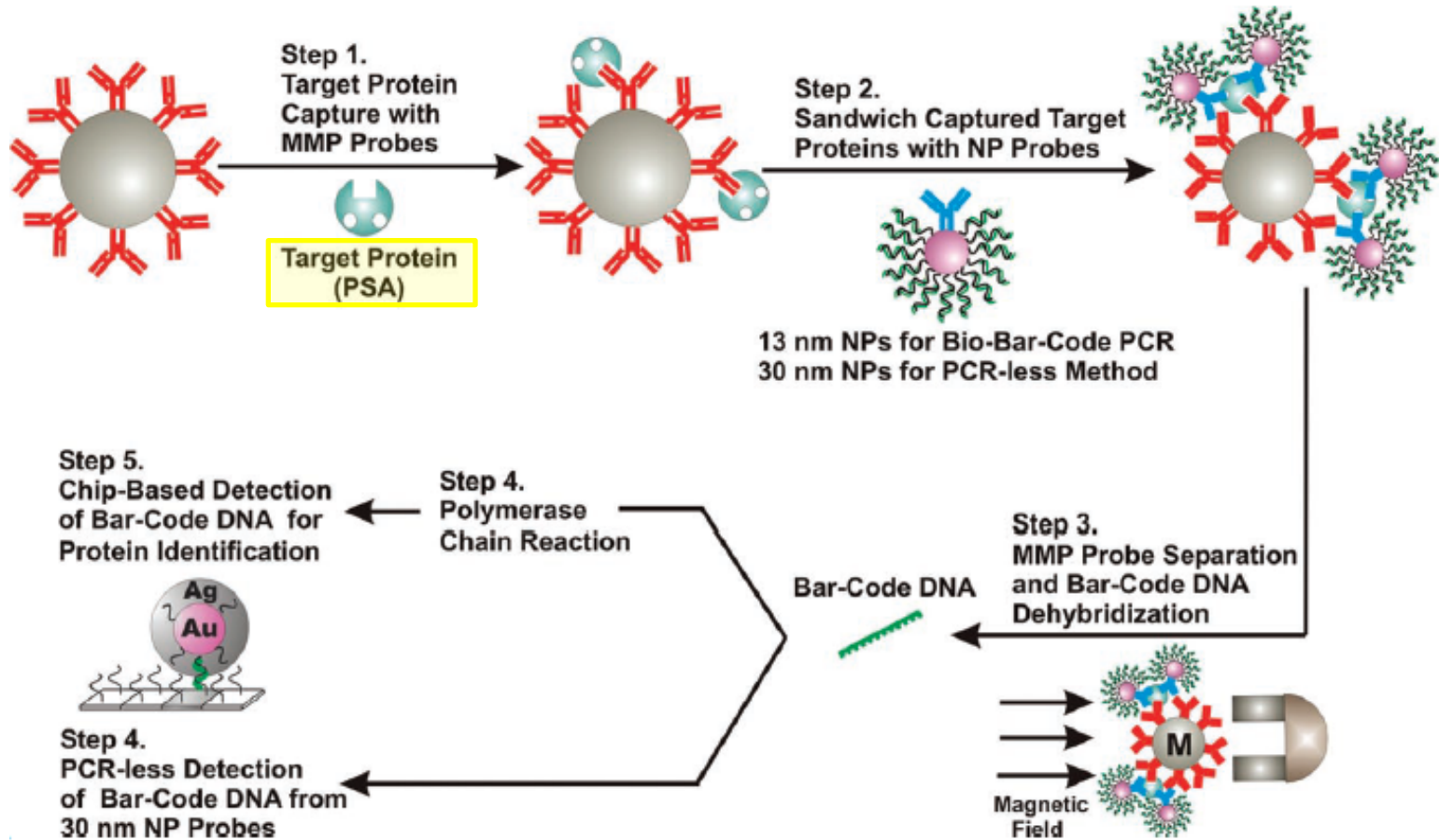


MMP: 1 μ m diameter



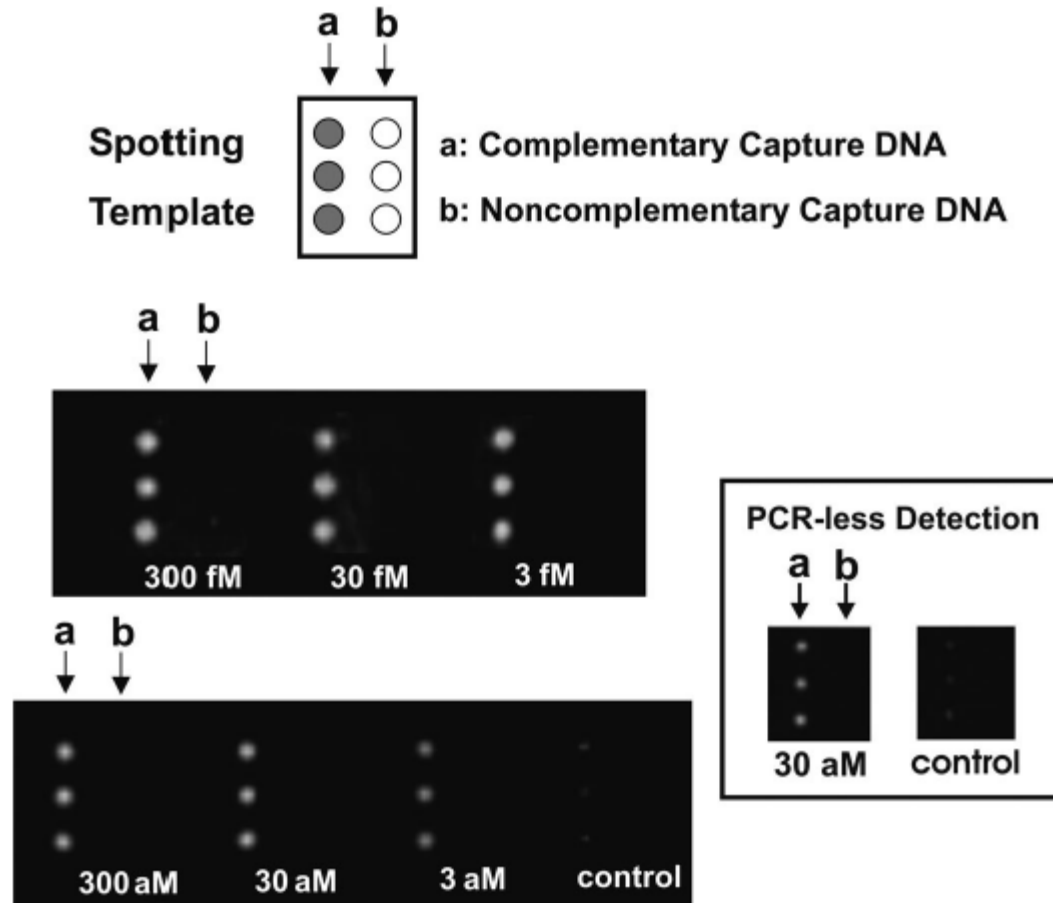
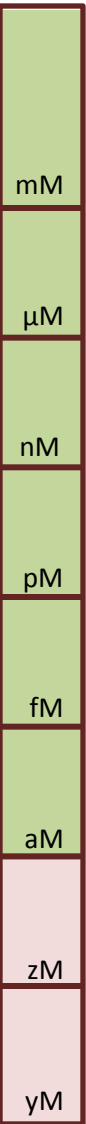


Detection of proteins using nanoparticle-based bio-bar codes





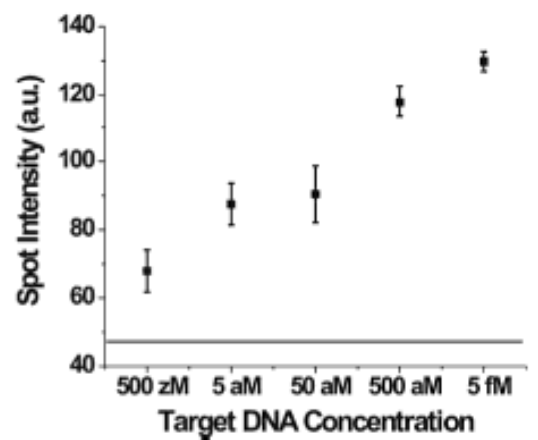
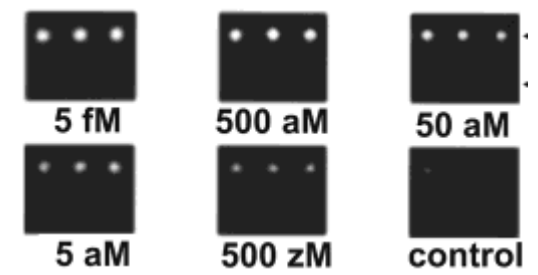
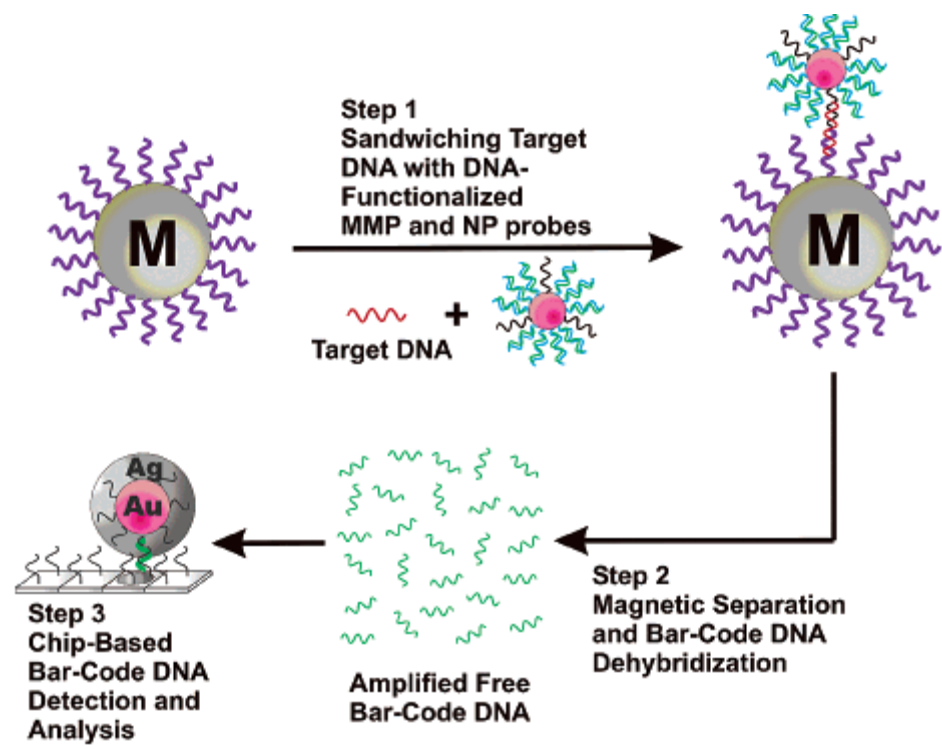
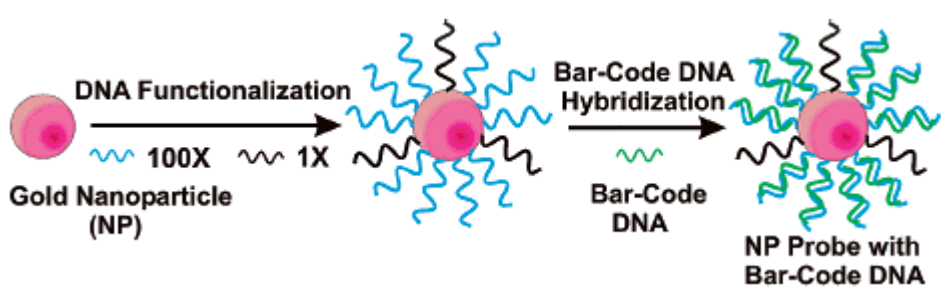
Signal production





Detection of DNA using nanoparticle-based bio-bar codes

- mM
- μ M
- nM
- pM
- fM
- aM
- zM
- yM





Concluding



Signal amplification

- catalysis
(difficult to reach extremely low concentrations – autocatalysis)
- multivalency
(essential)

Future

- cascade events
- beyond DNA, simple devices and readout



Acknowledgements

Funding:



European
Research
Council

239898



CM0703: Systems Chemistry
CM0905: Organocatalysis



READ-289723

